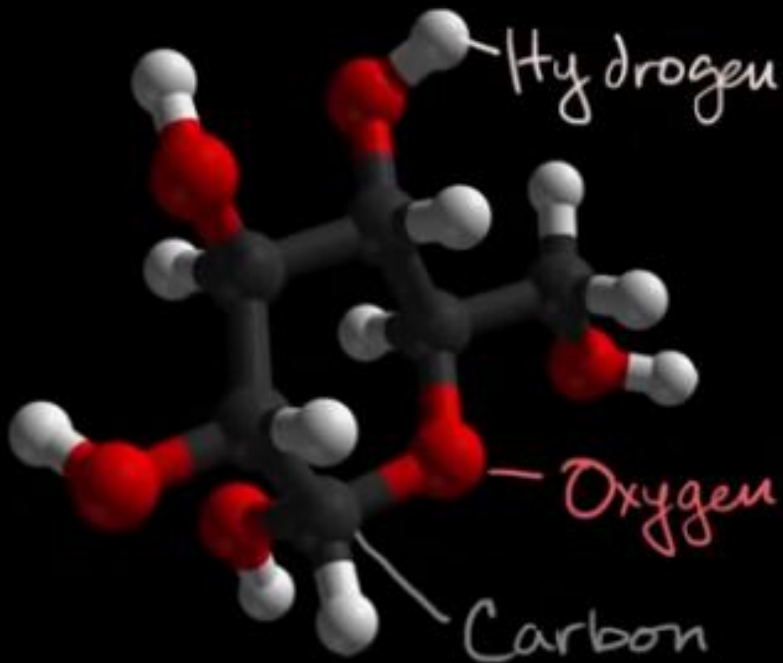
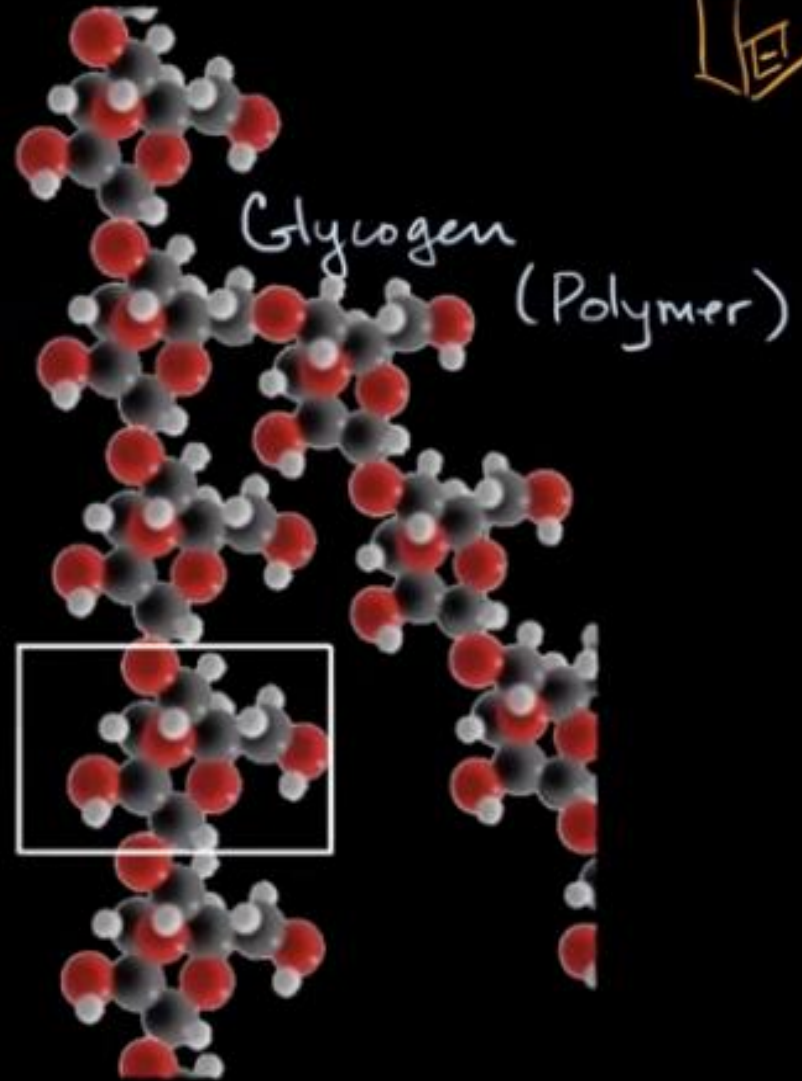


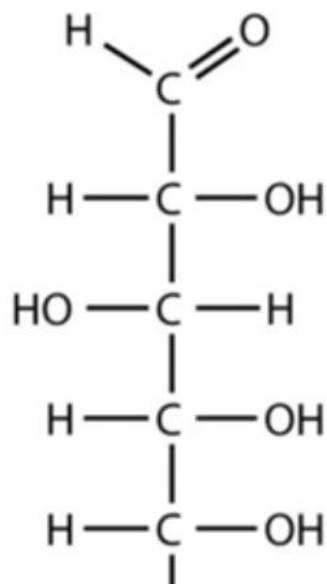
Carbohydrates



● Glucose
(Monomer)



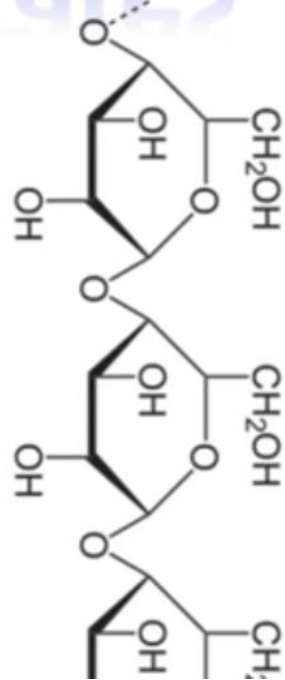
Classification of carbohydrates



Monosaccharides

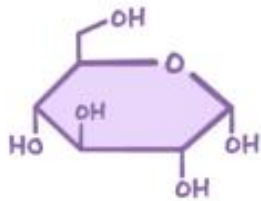
Oligosaccharides

Polysaccharides



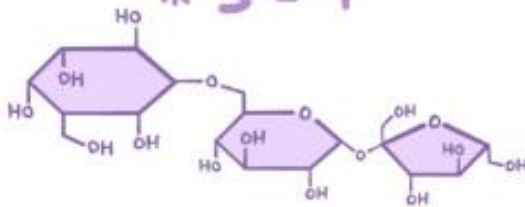
SUGAR ~ SACCHARIDES

ONE
MONOSACCHARIDE

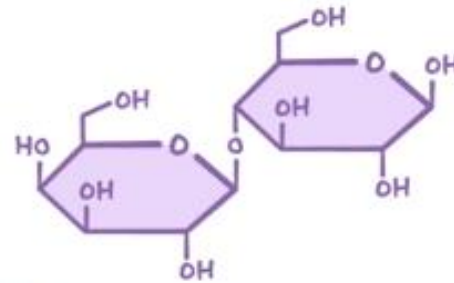


A FEW
OLIGOSACCHARIDE

* 3 - 9



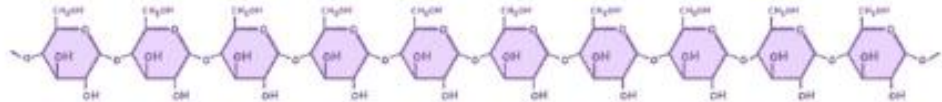
TWO
DISACCHARIDE



MANY

POLYSACCHARIDE

* 10 OR MORE



Monosaccharides

Simplest carbohydrates found in the nature



Monosaccharides

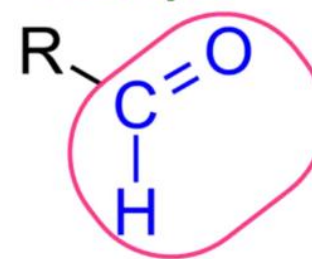
Aldoses

Ketoses

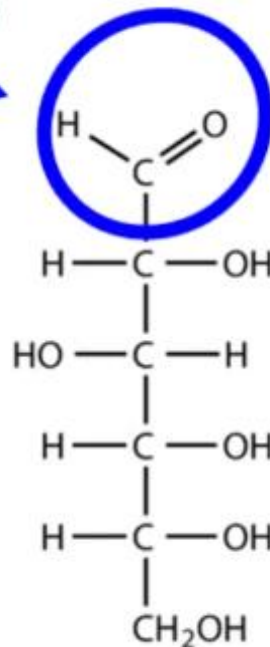


Aldehyde

Ketone

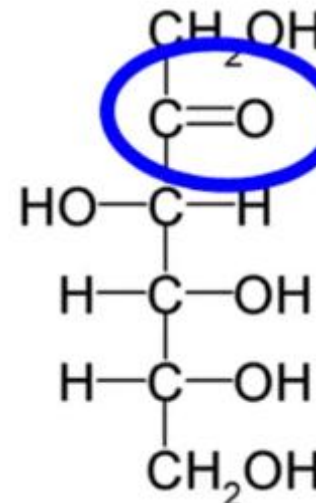


Aldehyde



Glucose

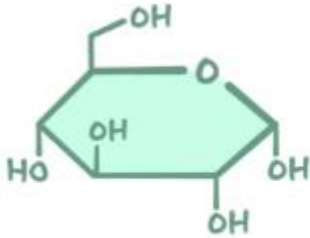
Ketone



Fructose

GLUCOSE

* MOST IMPORTANT *

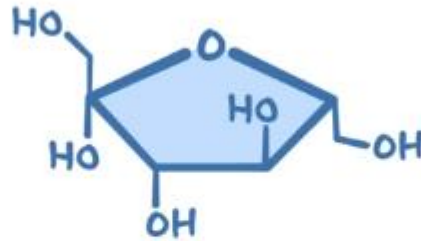


* ONE OF THE MAIN SOURCES OF CALORIES

* CAN CROSS **BLOOD-BRAIN** BARRIER

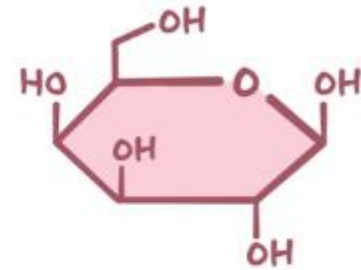
↳ **NOURISHES THE BRAIN**

FRUCTOSE

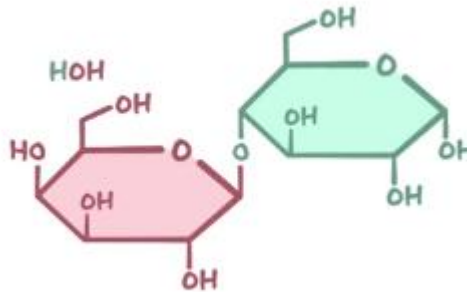


* COMMONLY FOUND IN **HONEY, FRUITS, & ROOT VEGETABLES**

GALACTOSE

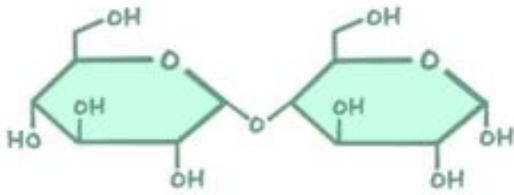


* **"MILK SUGAR"**



Glucose
+
Glucose

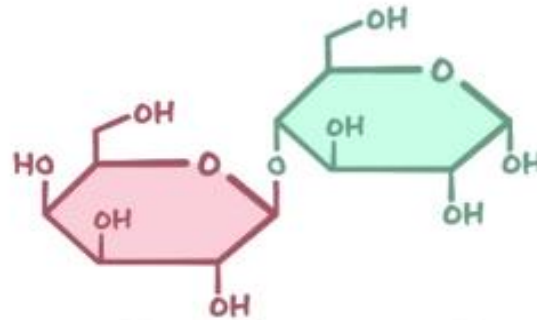
MALTOSE



* **FOUND IN MOLASSES**
↳ USED TO **FERMENT BEER**

Glucose
+
Galactose

LACTOSE

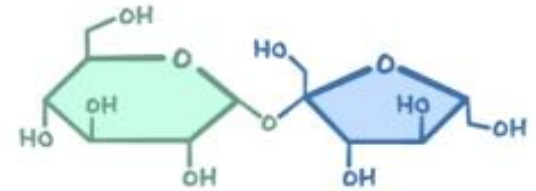


* **FOUND IN MILK OF MAMMALS**
↳ E.G. **COW & BREAST MILK**

Glucose
+
Fructose

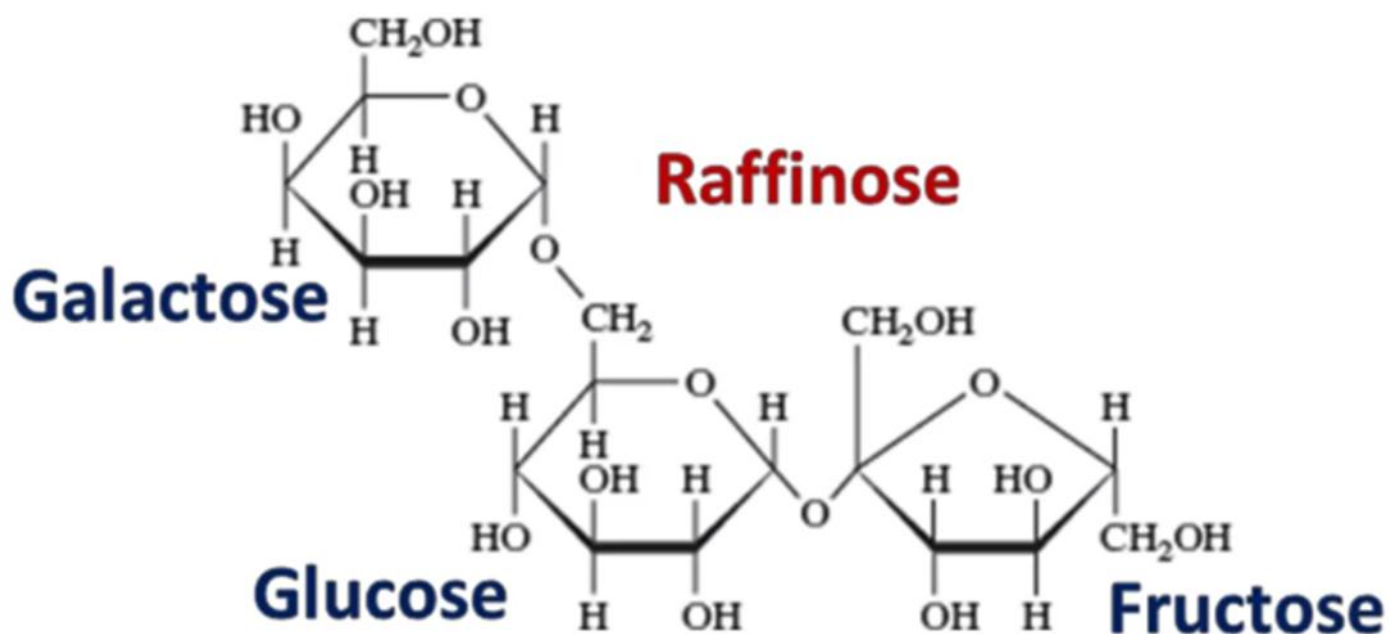
SUCROSE

~ "TABLE SUGAR" ~



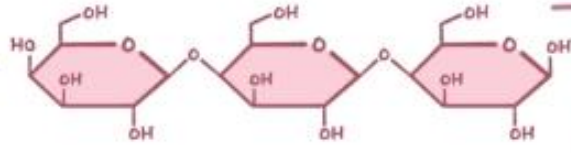
* **FOUND IN FRUITS & VEGETABLES**
↳ E.G. **SUGAR CANE & SUGAR BEETS**

Trisaccharide



COMPLEX CARBOHYDRATES

OLIGOSACCHARIDES ~

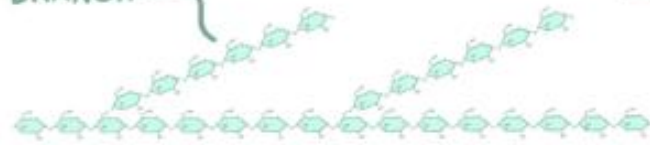


GALACTO-OLIGOSACCHARIDE
* IN SOYBEANS

POLYSACCHARIDES ~

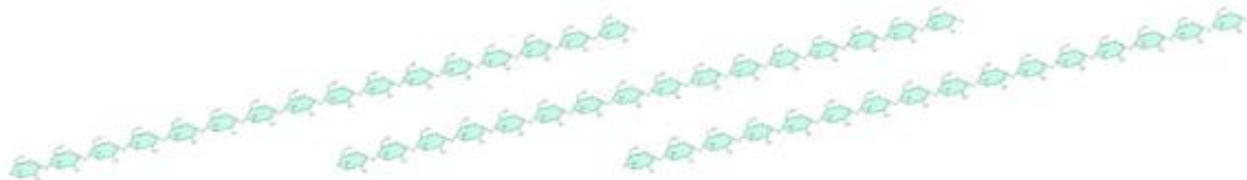
* MOST ABUNDANT
CARBOHYDRATE IN FOOD

BRANCH



STARCHES

- * INTESTINAL ENZYMES CAN BREAK DOWN
- * IN RICE, POTATOES WHEAT & MAIZE
- * NOT SWEET



DIETARY FIBERS

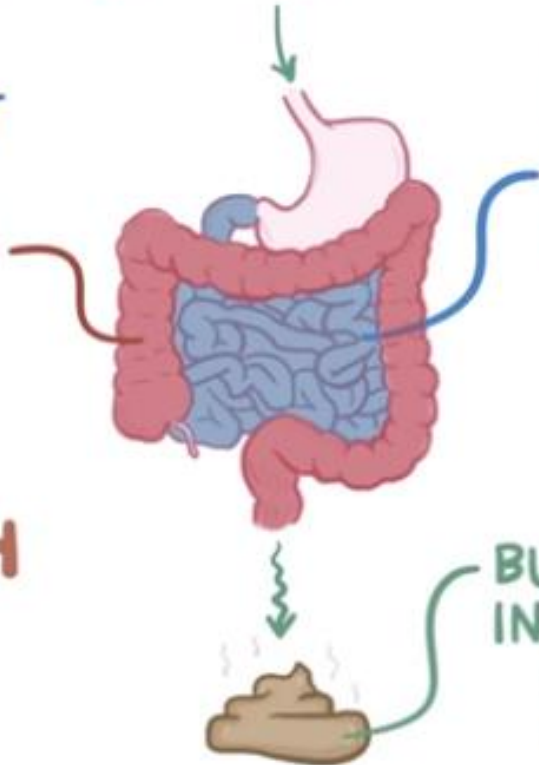
- * INTESTINAL ENZYMES CANNOT BREAK DOWN

DIETARY FIBERS



BROKEN DOWN A BIT
BY BACTERIA IN
LARGE INTESTINE

* BETA-GLUCAN
GOOD FOR
HEART HEALTH



PASSES THROUGH
SMALL INTESTINE
UNDIGESTED

SLOWS ABSORPTION
FOR SIMPLE SUGARS

MAINTAIN
BLOOD GLUCOSE

BULK MATTER
IN STOOL

- * INCREASED WEIGHT
- * DECREASED CONSTIPATION

Polysaccharides

Homo polysaccharide

Hetero polysaccharide

Polymer of same monosaccharide units

Polymer of different monosaccharide units

Homo polysaccharide

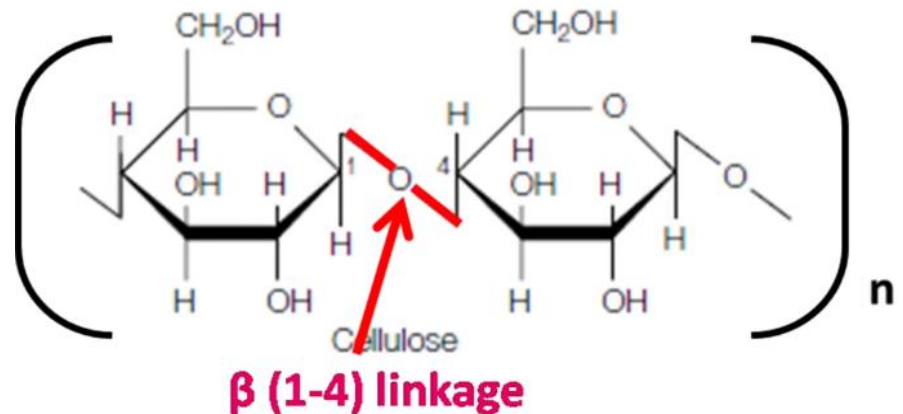
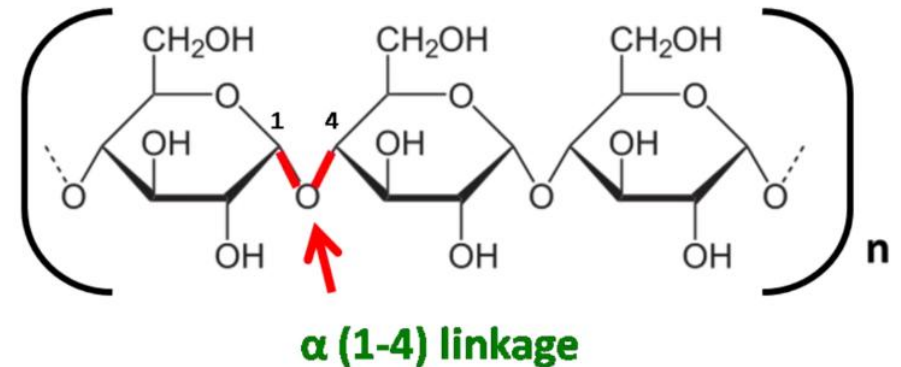
α (1-4) linkage

Starch

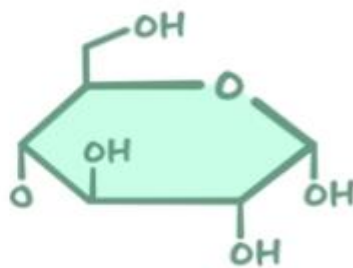
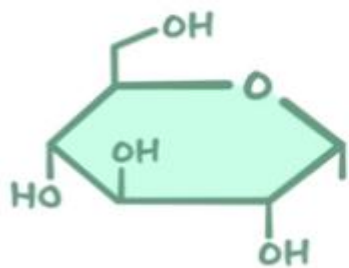
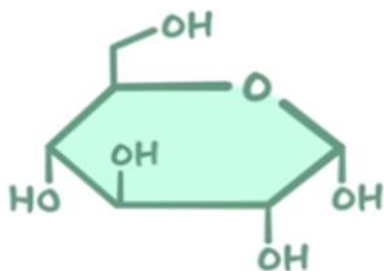
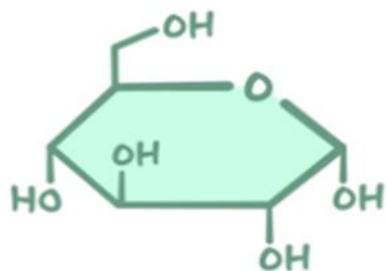


β (1-4) linkage

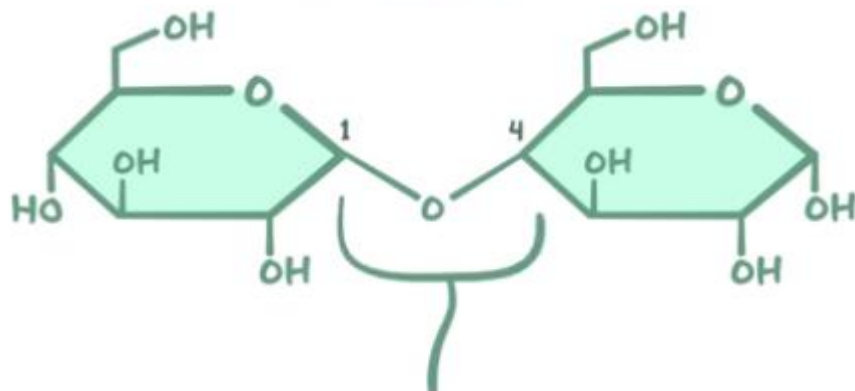
Cellulose



GLYCOSIDIC BONDING



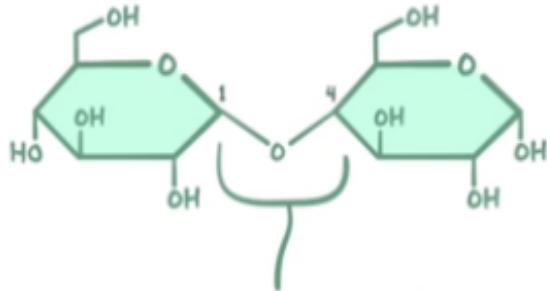
~ MALTOSE ~



α 1-4-GLYCOSIDIC BOND

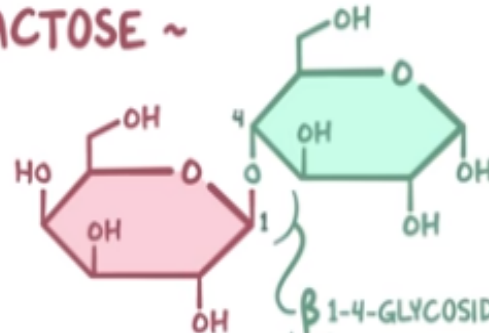
GLYCOSIDIC BONDING

~ MALTOSE ~



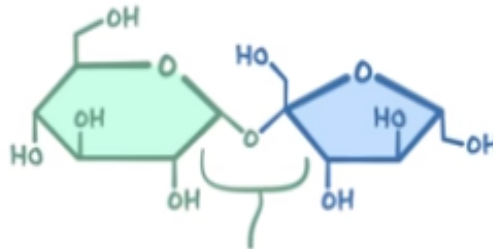
α 1-4-GLYCOSIDIC BOND
LINED UP

~ LACTOSE ~



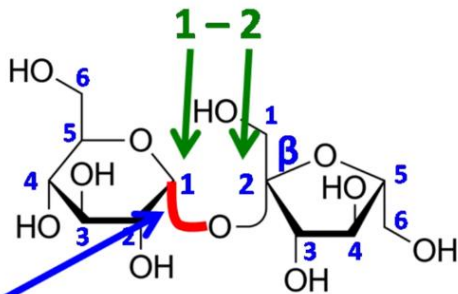
β 1-4-GLYCOSIDIC BOND
STACKED

~ SUCROSE ~



α 1-2-GLYCOSIDIC BOND

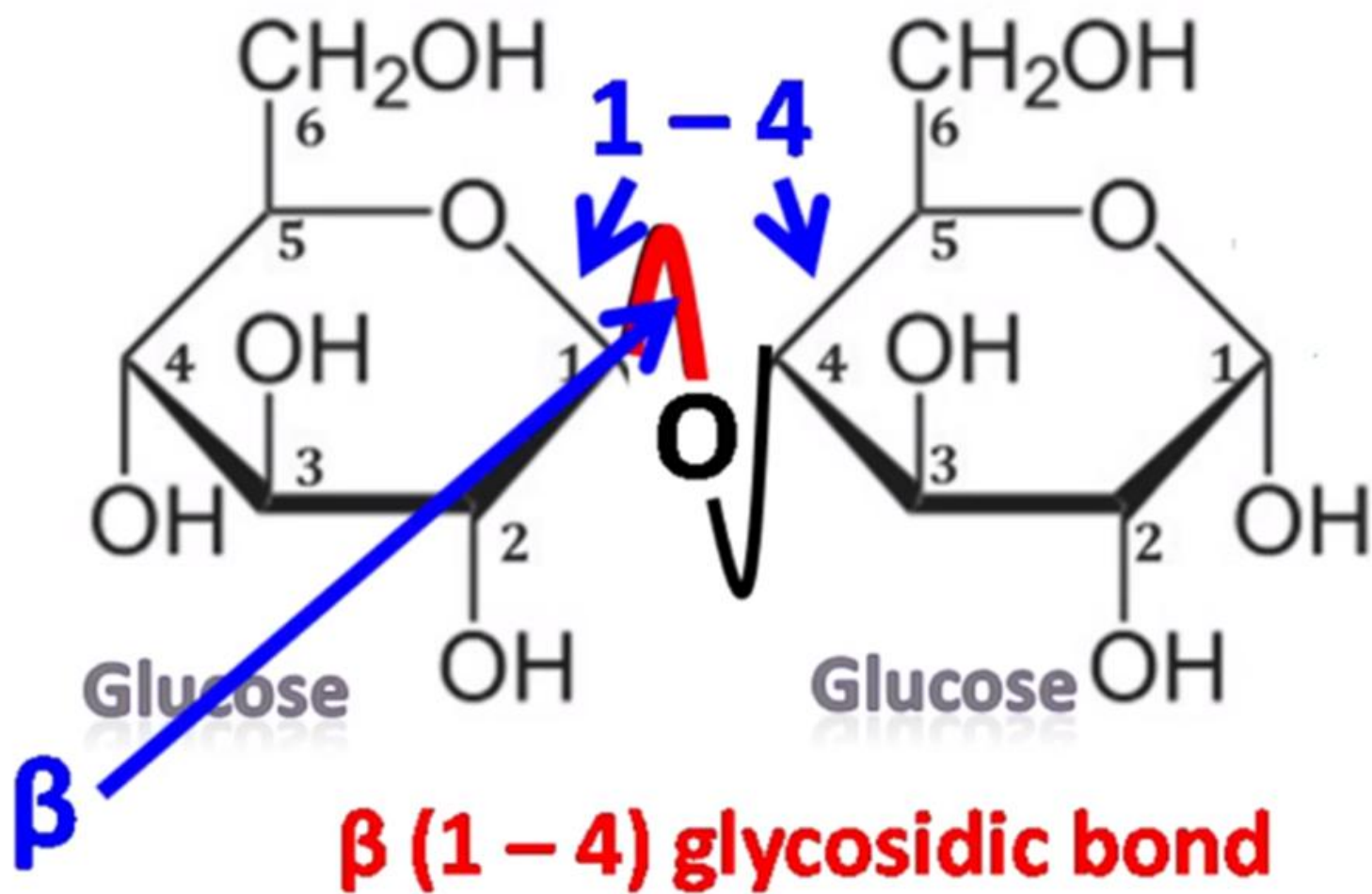
Sucrose



Glucose Fructose

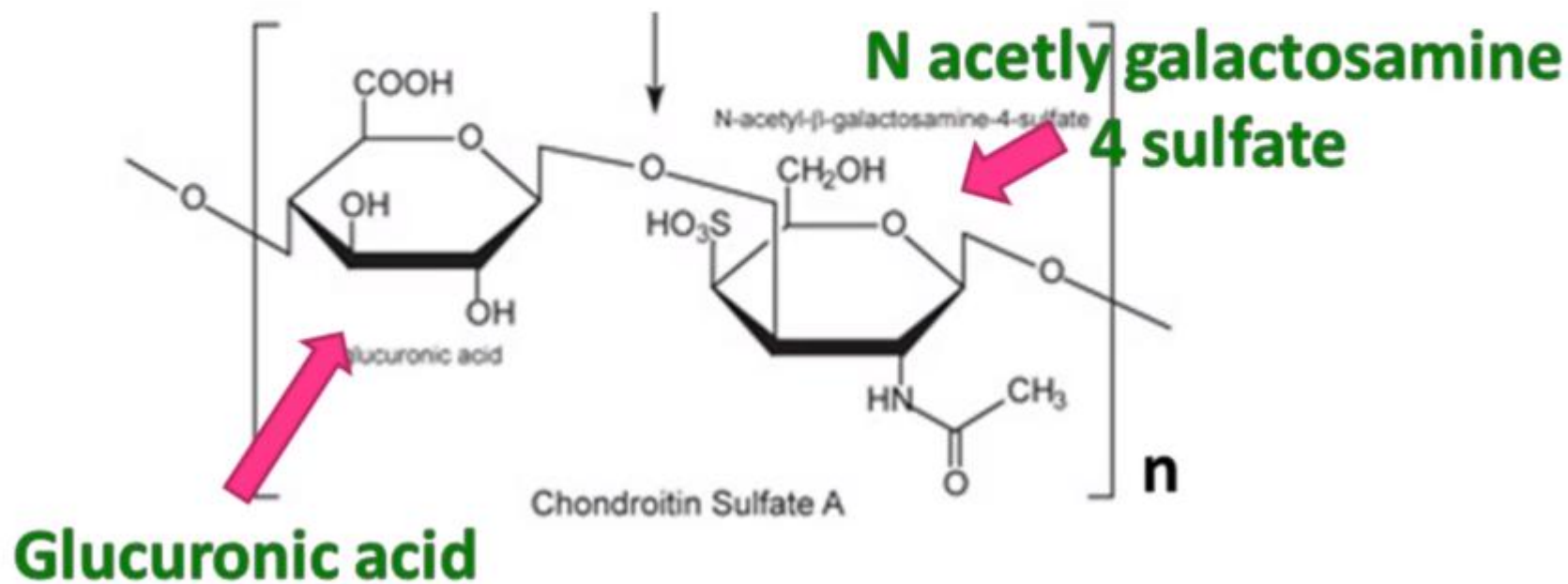
α (1 - 2) glycosidic bond

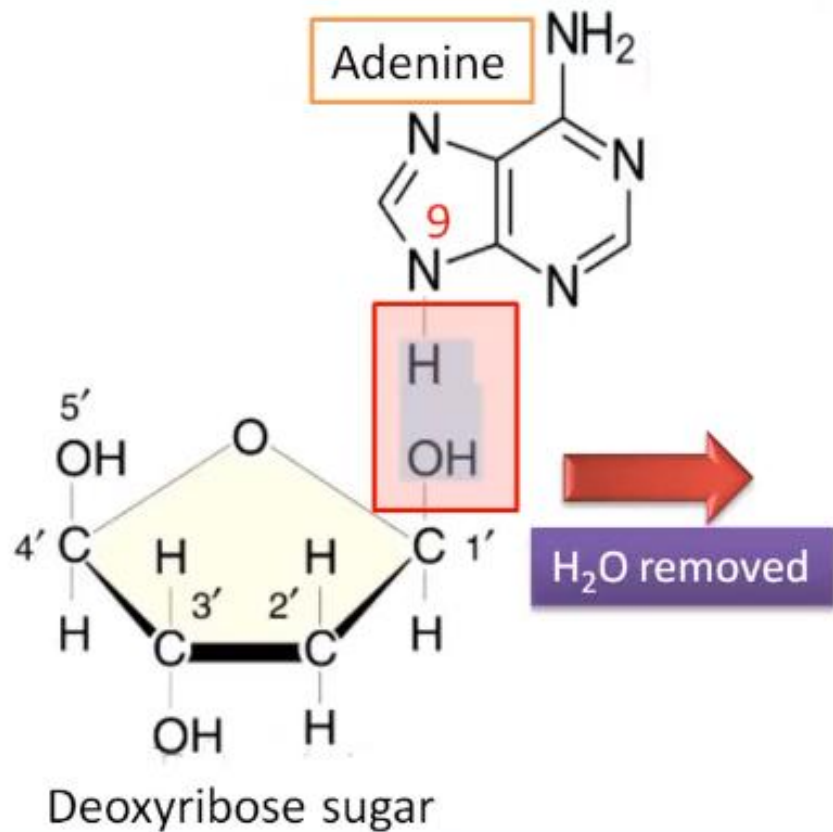
Cellulose



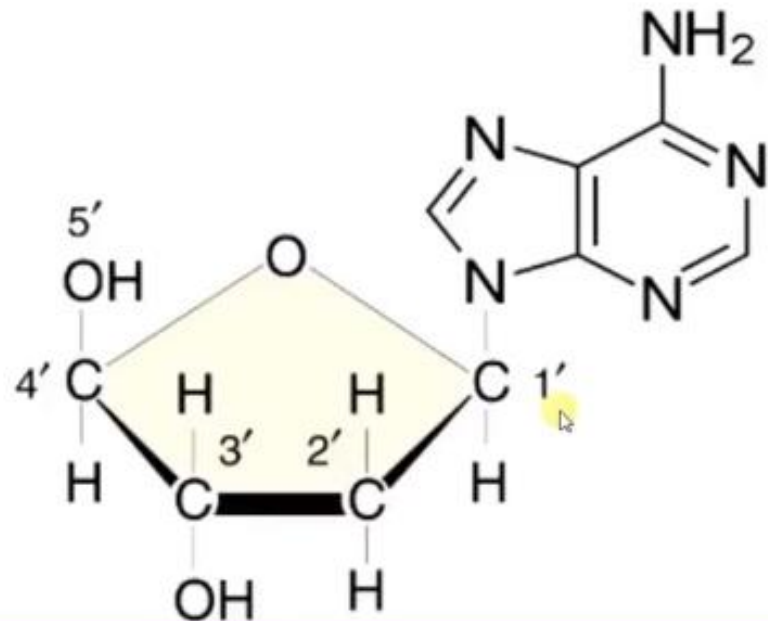
Hetero polysaccharide

Chondroitin 4 sulfate

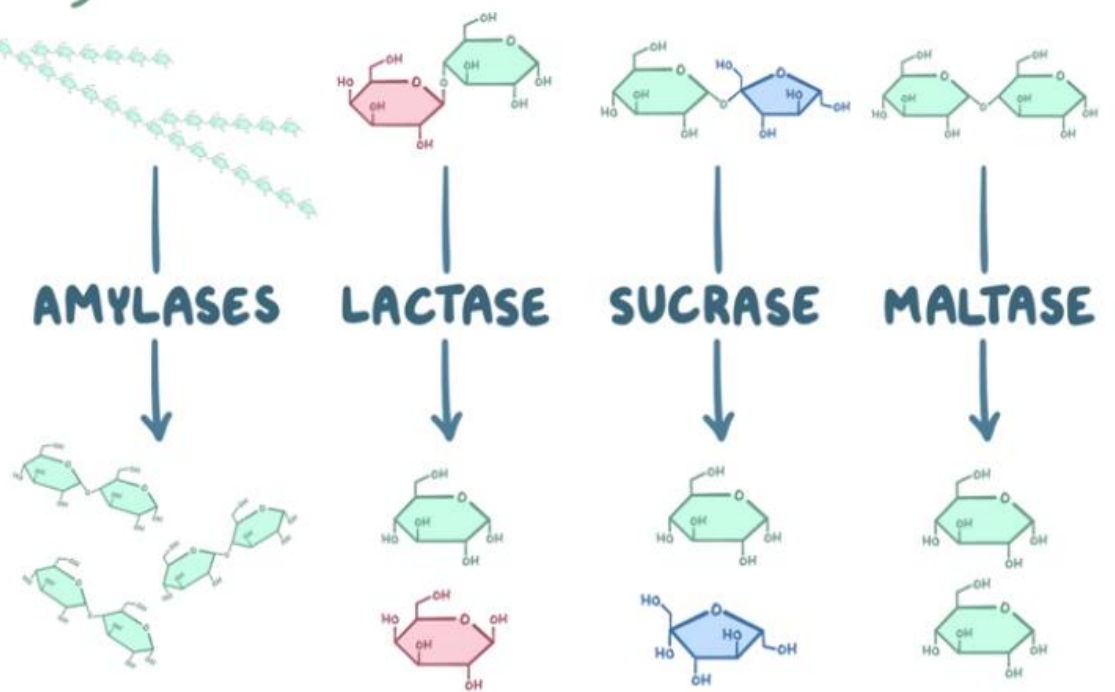
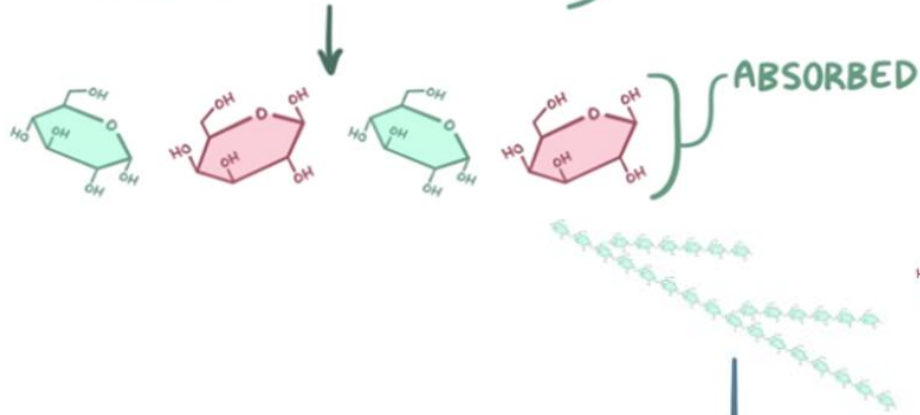
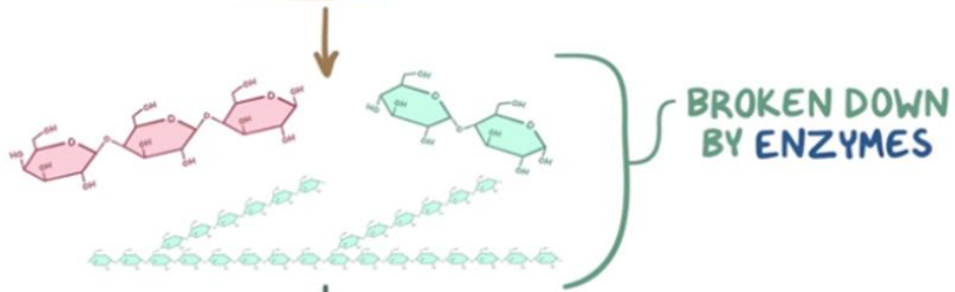


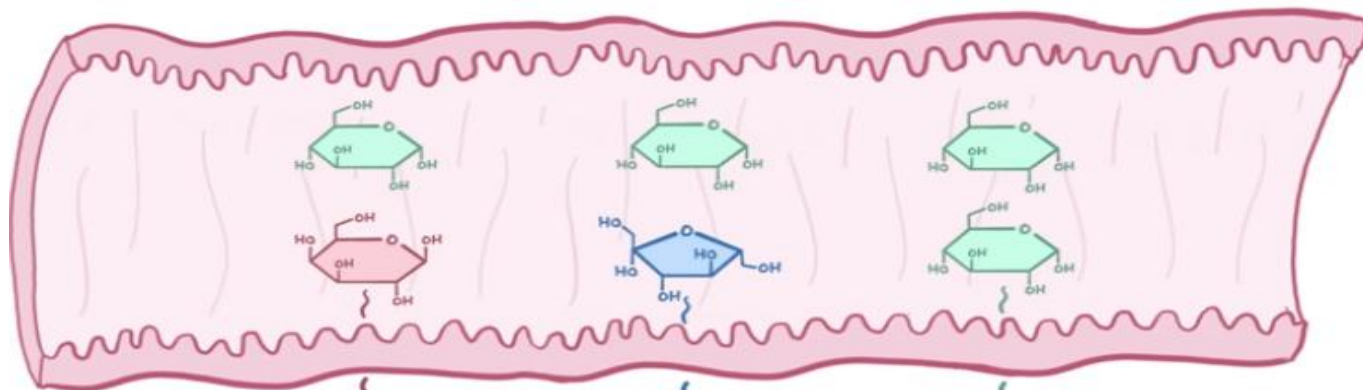


Glycosidic Bond in DNA



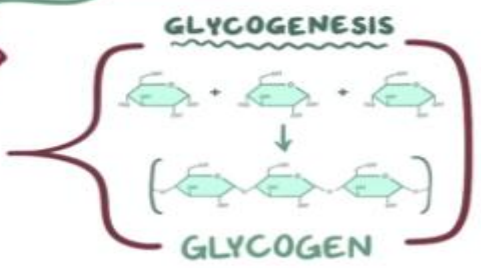
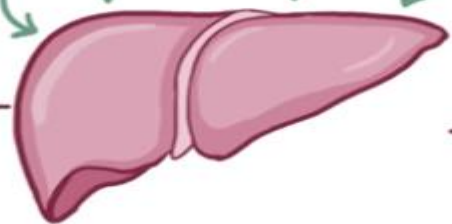
In DNA, Glycosidic Bond, **N-C linkage** connects Sugar with Nitrogenous base.
The **C1** of deoxyribose forms **β -N-Glycosidic bond**



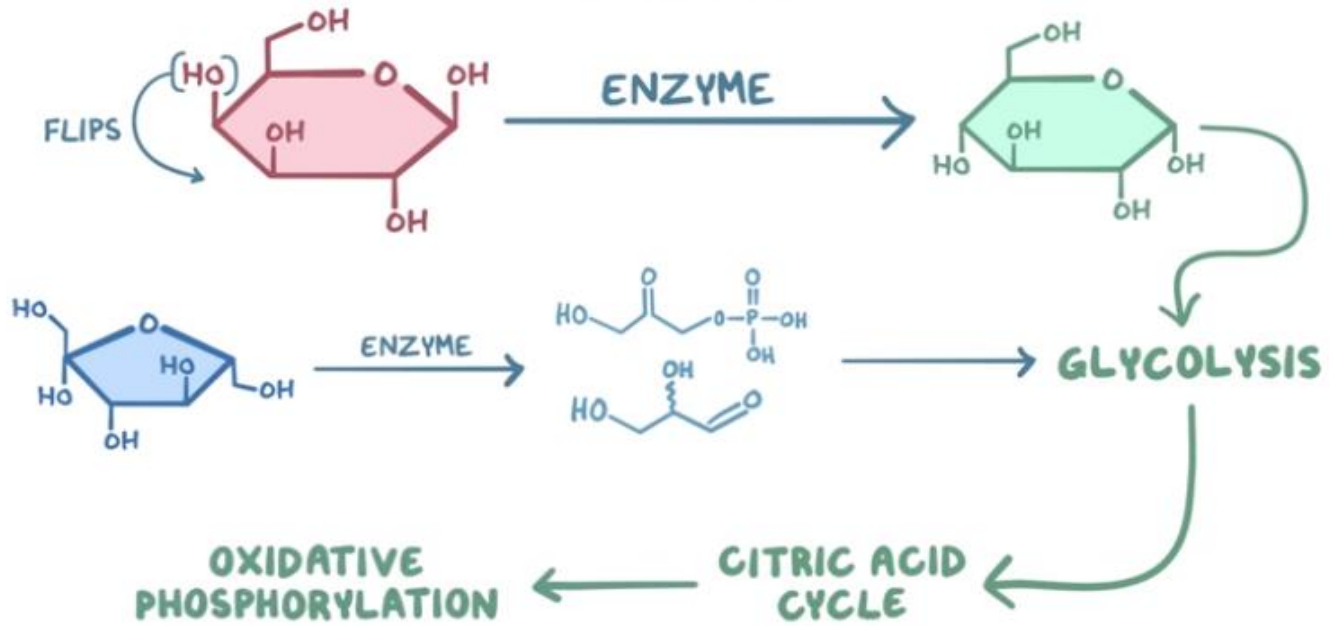


INSULIN

- * MOVES GLUCOSE INTO ALL CELLS FOR ENERGY
- * STIMULATES THE **LIVER**
- * PROMOTES FAT & PROTEIN SYNTHESIS



(IN LIVER)



HONEY



POTATOES



SIMPLE SUGARS

STARCHES

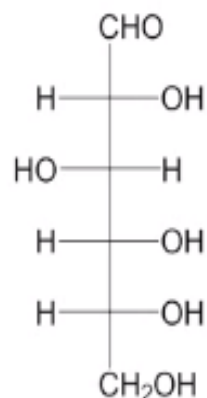
MONOSACCHARIDES

STORAGE

* IMMEDIATE ENERGY *

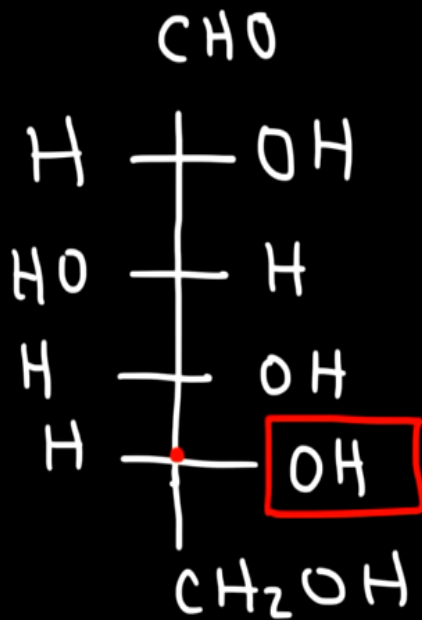
Carbohydrate Structure

- Polyhydroxylated carbonyl compounds (usually aldehydes or ketones)
- General formula: $C_n(H_2O)_n$ (one unsaturation)



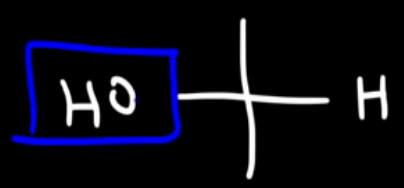
Fischer projection
(open chain view)



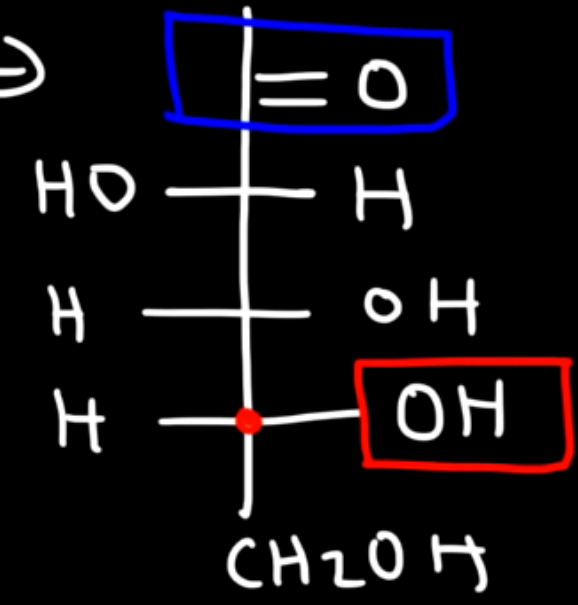


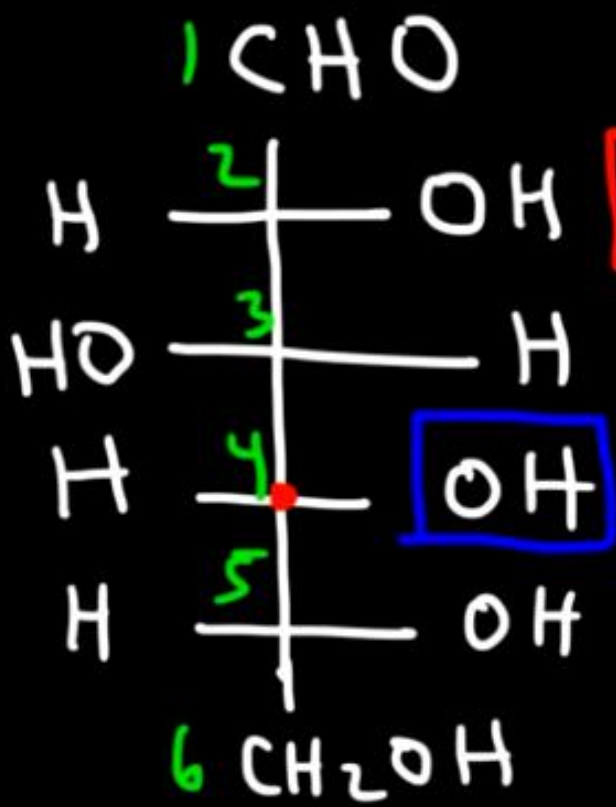
D-Glucose

L-Glucose



D-Fructose CH_2OH

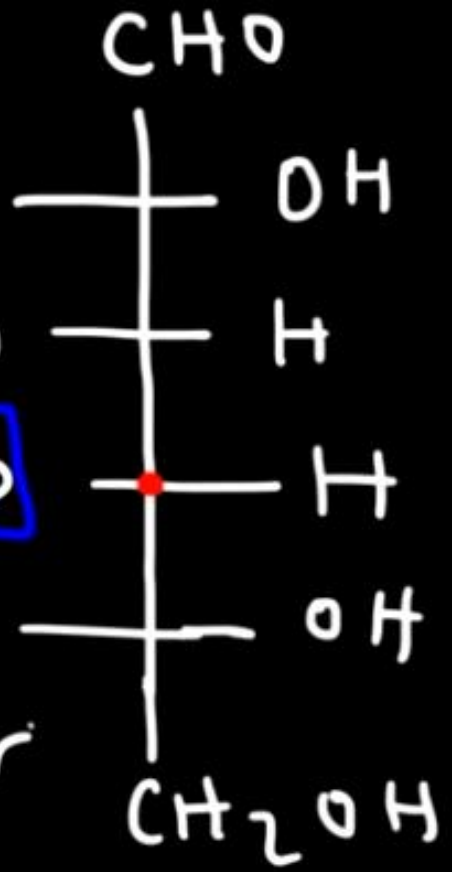




epimers

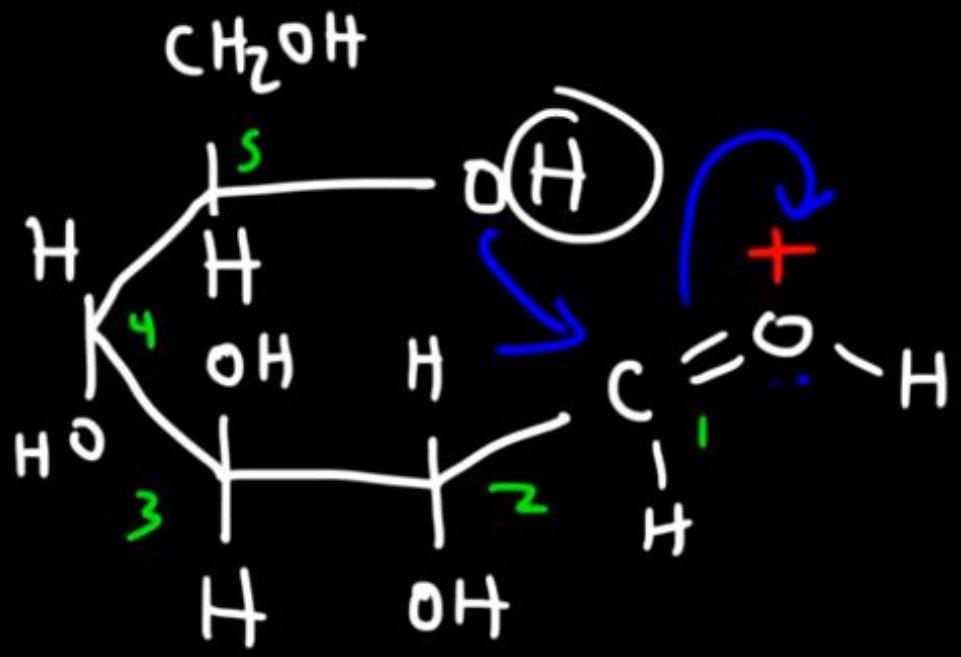
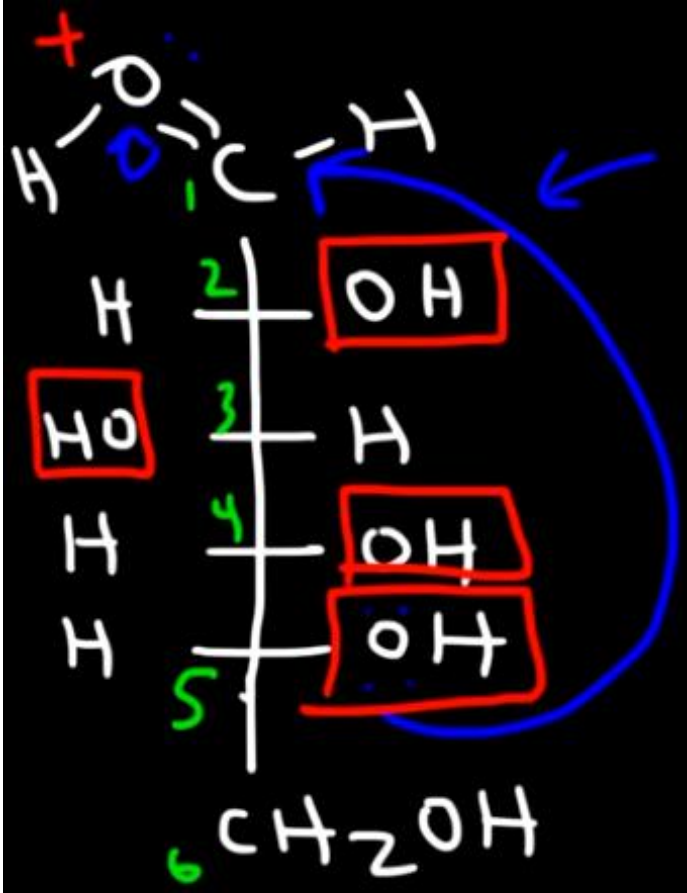


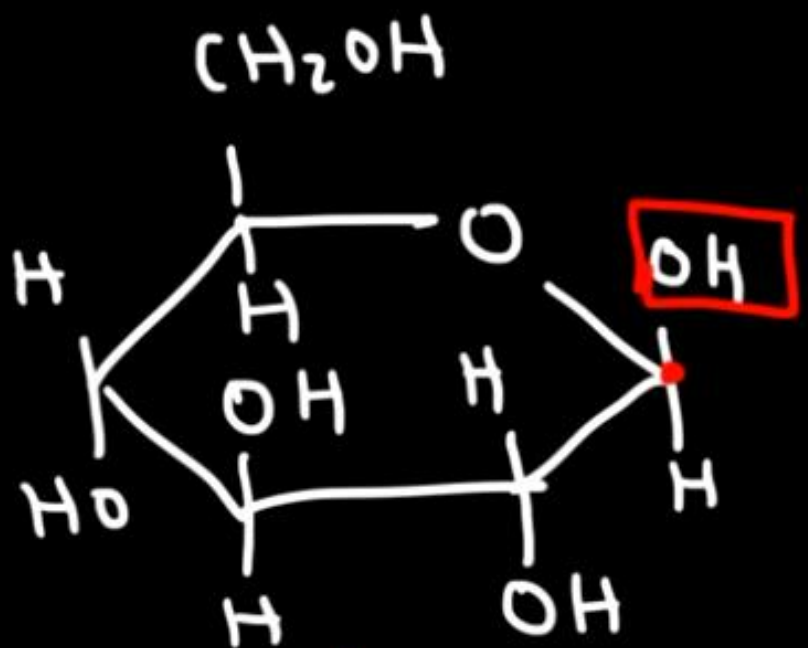
C-4 epimer



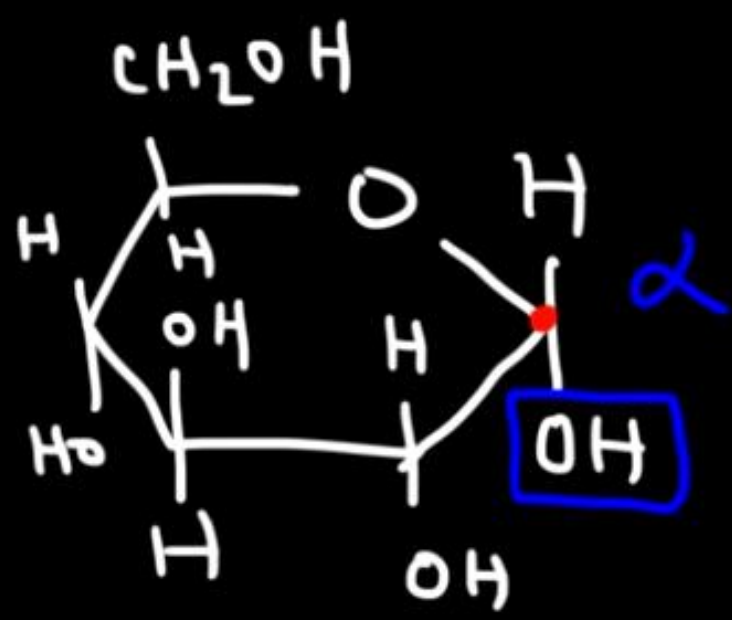
D-Glucose

D-Galactose





B
64%



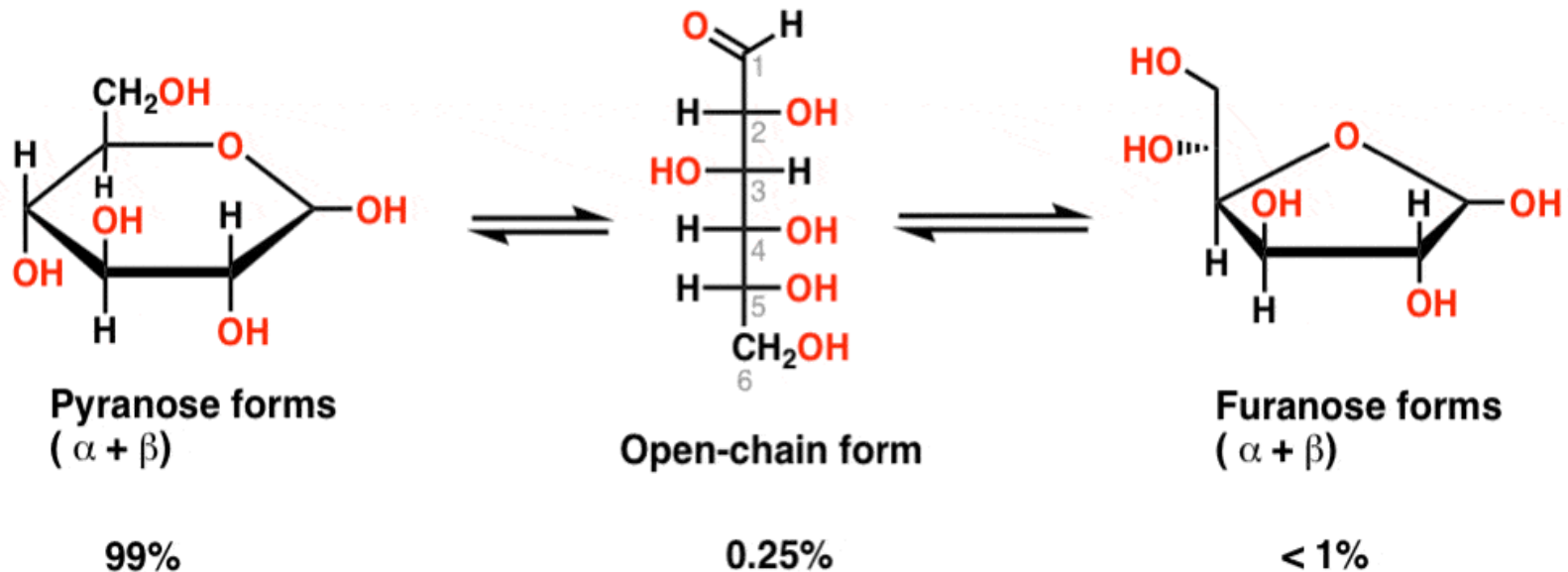
36%

Summary: Pyranoses, Furanoses, and Ring-Chain Tautomerism

Sugars exist in equilibrium between their open-chain and various closed-chain forms. (This is called "ring-chain tautomerism")

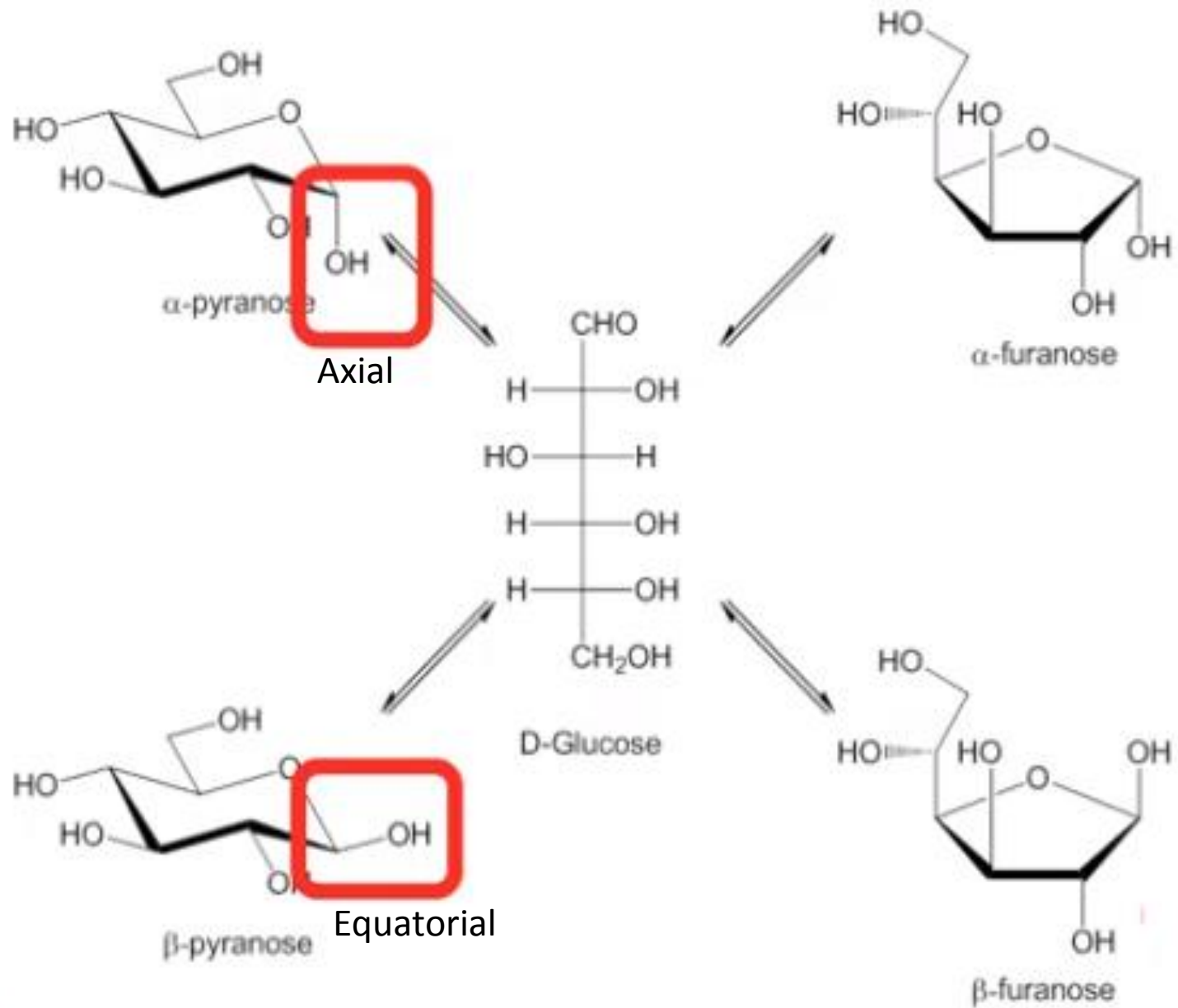
Particularly important for hexoses (e.g. glucose) and pentoses (e.g. ribose)

e.g. D-glucose (below)

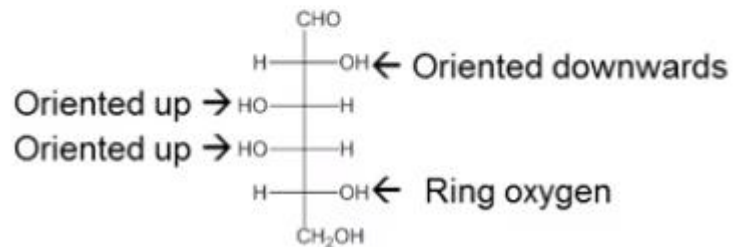
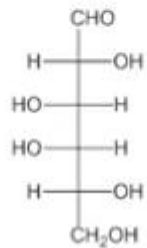
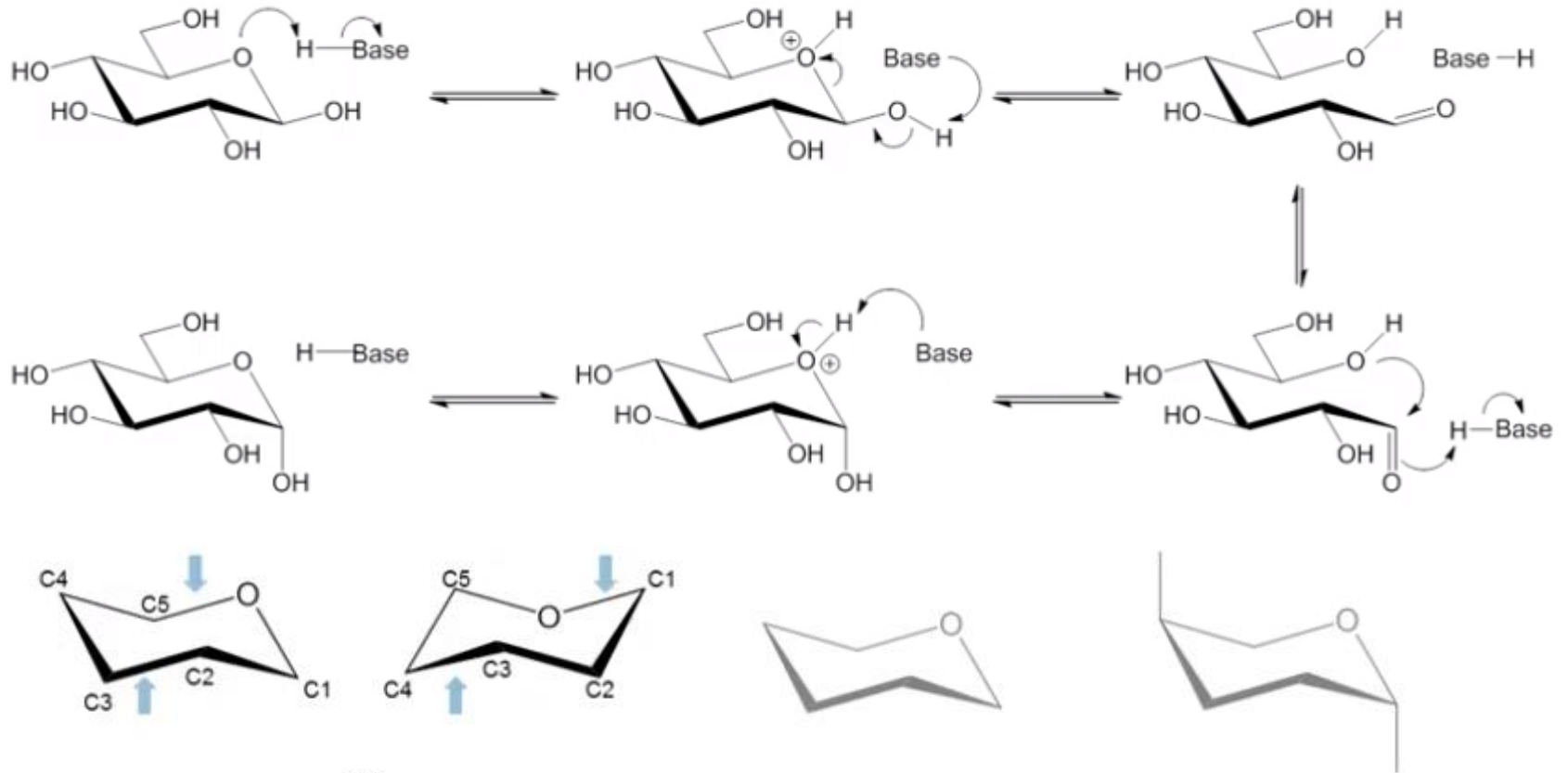


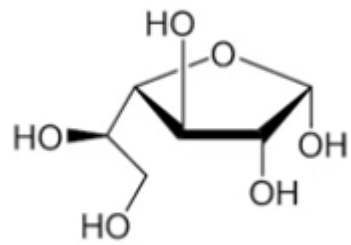
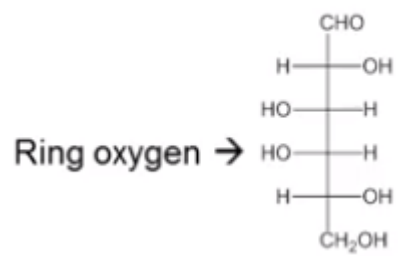
The six-membered cyclic form is generally referred to as the "pyranose" form, and the five-membered cyclic form is called the "furanose" form

Closure of the ring creates a chiral center at C-1, resulting in two diastereomers (sometimes called "anomers") - the alpha (α) and beta (β) forms

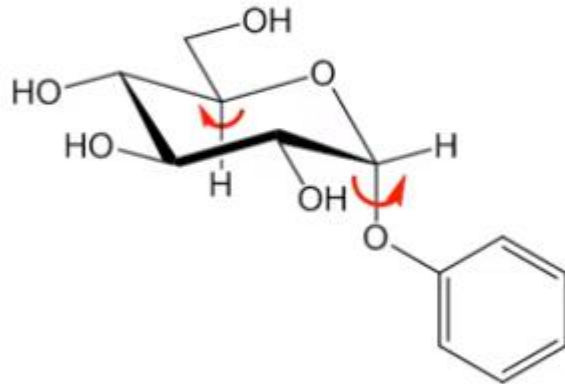


Mutarotation

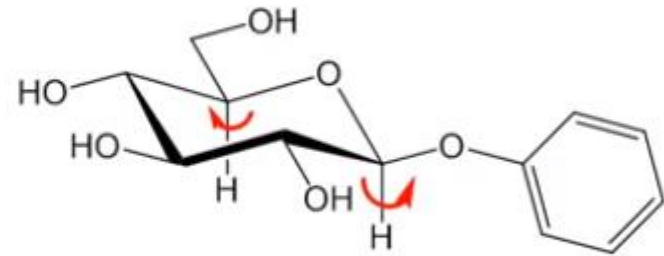




Glycosides



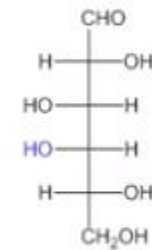
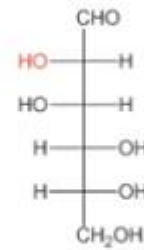
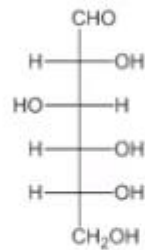
Oxygen @ C-5 is on the right
Oxygen @ C-1 is on the right
Configuration is α



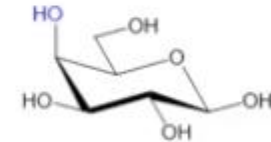
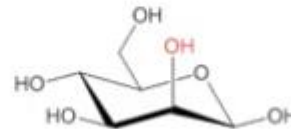
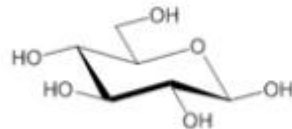
Oxygen @ C-5 is on the right
Oxygen @ C-1 is on the left
Configuration is β

Common Sugars

Fischer projection:



Line drawing:



Name:
(Abbreviation)

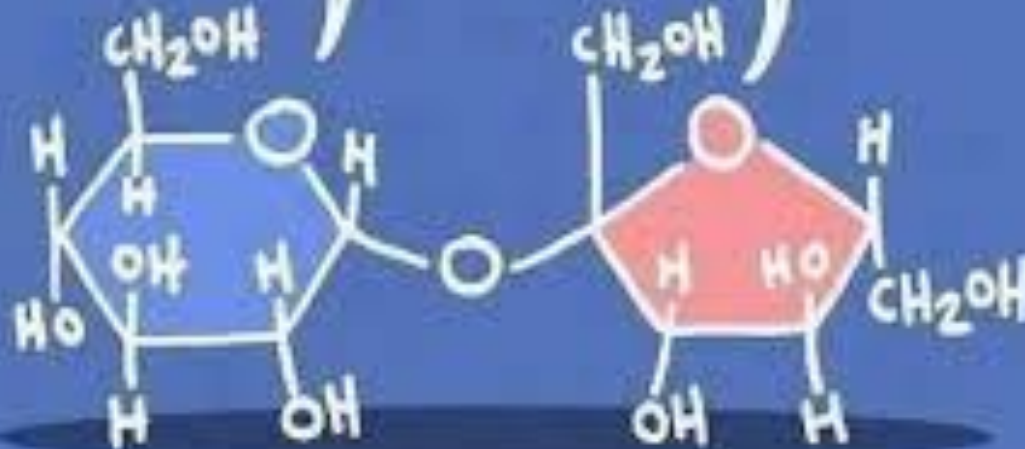
β -D-glucose (Glc)

β -D-mannose (Man)

β -D-galactose (Gal)

You are
so sweet

You sweeten
my life

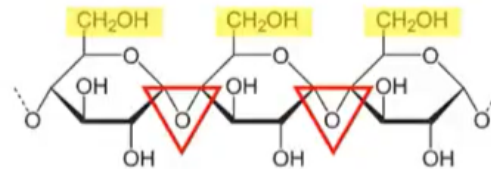


1. Which polysaccharide makes up plant cell walls?
2. Which monosaccharide makes up this polysaccharide?
3. Is this polysaccharide branched or unbranched?

Alpha (α) vs. Beta (β) glycosidic linkages

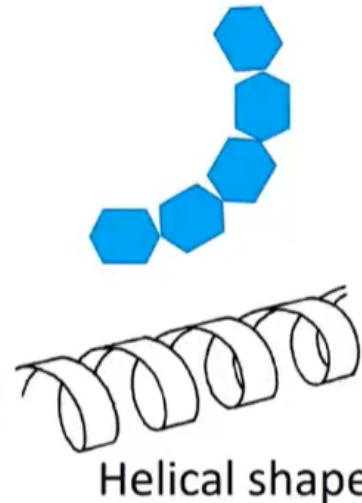
- Alpha (α)

- Create **HELICAL** shaped polysaccharide
- Humans **CAN** digest/break them apart
 - Our guts (digestive systems) make enzymes that hydrolyze α linkages



<https://upload.wikimedia.org/wikipedia/commons/thumb/e/ec/Amylose3.svg/800px-Amylose3.svg.png>

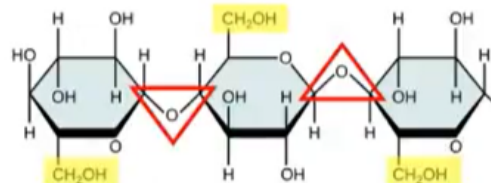
https://commons.wikimedia.org/wiki/File:Hand-drawn_helix_ribbons_at_various_angles.jpg



Helical shape

- Beta (β)

- Create straight chain shaped polysaccharides
- Humans **CANNOT** digest/break them apart
 - Our guts do NOT make any enzymes that can hydrolyze β



https://upload.wikimedia.org/wikipedia/commons/b/ba/Figure_03_02_07.jpg



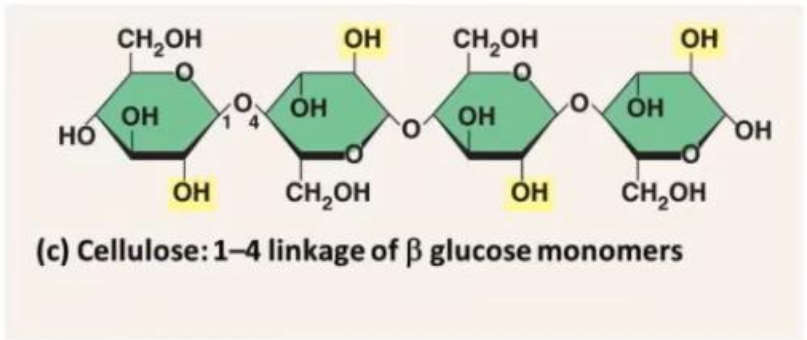
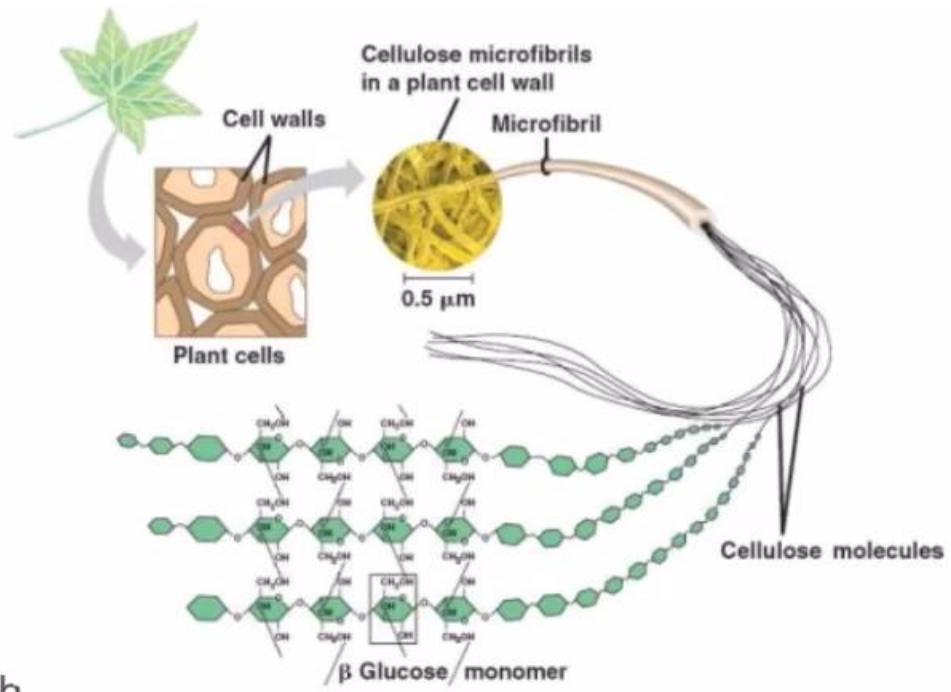
Straight chain shape

Cellulose - structure

Cellulose is made up of β -glucose monomers linked by β 1-4 glycosidic bonds.

Cellulose does not branch.

The monomers are packed tightly as extended long chains held adjacent to each other by hydrogen bonding. This gives cellulose its rigidity and strength.



In cellulose, every β -glucose monomer is "upside down" with respect to its neighbors.

Cellulose - *function*

The cell wall of plants is mostly made of cellulose; this provides structural support to the cell.

Fun fact: wood and paper are mostly cellulose!

Cellulose!



Cellulose - *function (just read)*

The β 1-4 linkage found in cellulose cannot be broken down by human digestive enzymes.

The cellulose we eat is called "fiber" in every day lingo.

Fiber moves through the digestive tract undigested and then is egested from the body (poop).

Fiber has an important function of maintaining health of the digestive tract.



Why can't humans digest cellulose?

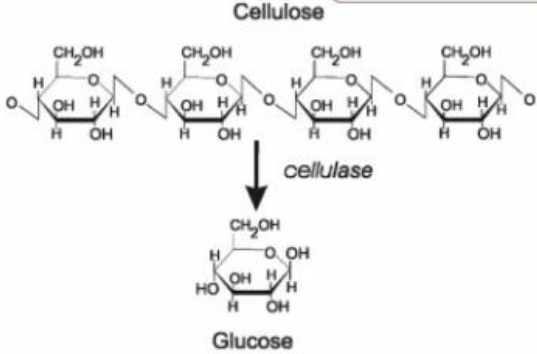
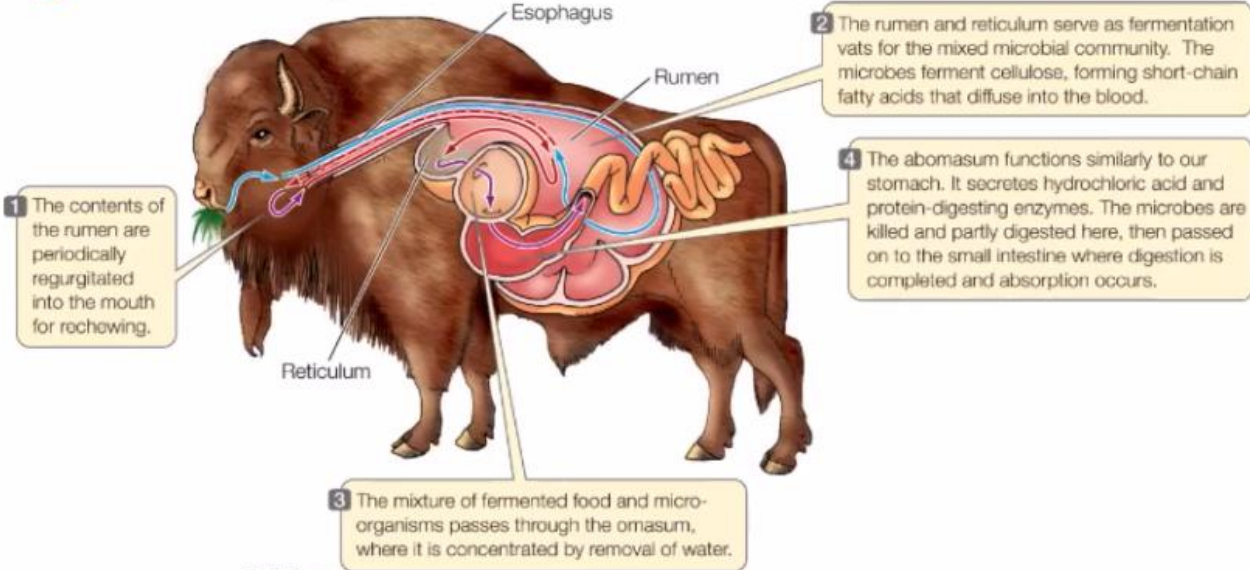
- We don't produce ***Cellulase***
- Ruminants, like cows, have symbiotic bacteria in their stomachs



Cellulose - *function (just read)*

Herbivores (such as cows, koalas, buffalos, and horses) have specialized bacteria in their digestive tracts that digest cellulose and use it as a food source.

In these animals, bacteria in the rumen secrete the enzyme cellulase. Cellulase breaks down cellulose into glucose monomers that can be used as an energy source by the animal.



Starch - structure

Starch is made up of α -glucose monomers.

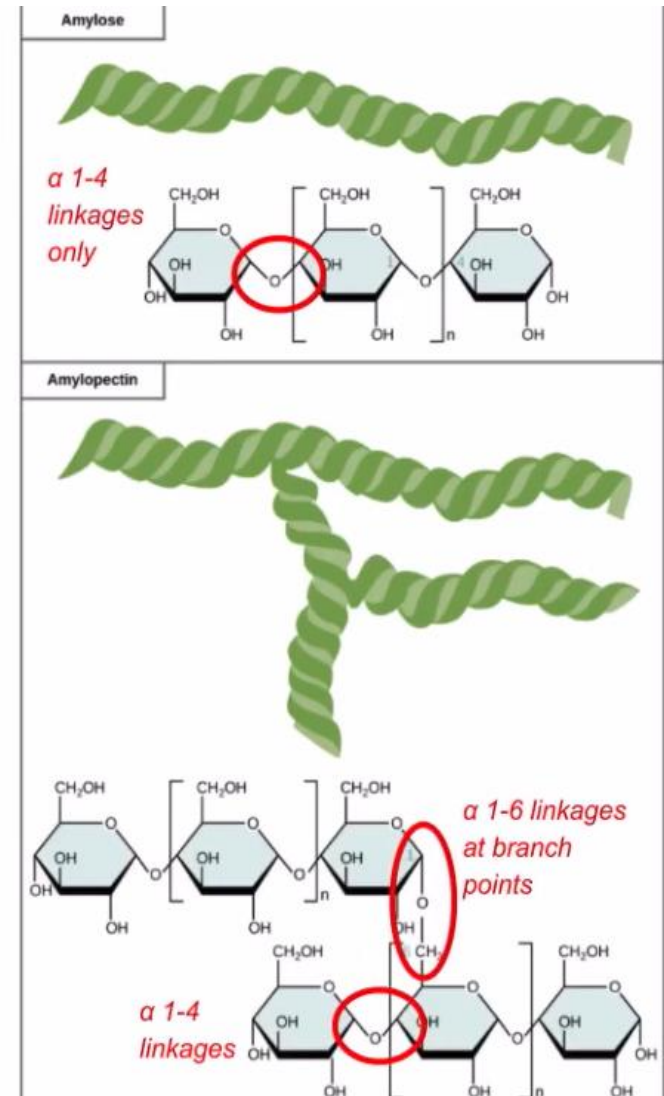
There are two types of starch: amylose and amylopectin (both polymers of α -glucose).

Amylose is starch formed by **unbranched** chains of glucose monomers (only α **1-4*** linkages)

Amylopectin is a **branched** polysaccharide (α **1-4*** linkages and α **1-6** linkages at the branch points).

**The numbers 1-4 and 1-6 refer to the carbon number of the two monosaccharides that have joined to form the bond.*

- α -glucose molecules joined by α 1-4 glycosidic bonds with α 1-6 branches every 20-30 monomers



Starch - *function*

Starch* is the storage form of carbohydrates in plants.

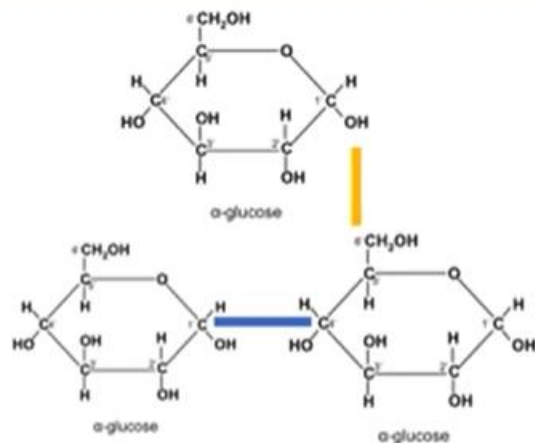
Plants are able to synthesize glucose in photosynthesis, and the excess glucose, beyond the plant's immediate energy needs, is stored as starch in different plant parts, including roots and seeds.

The plants form the basis of the food chain. Animals eat the plants as a source of carbohydrates.



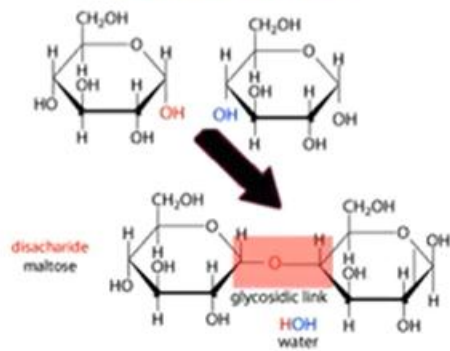
**Both amylose and amylopectin forms of starch have the same function*

Starch: Amylose and Amylopectin



Picture 1

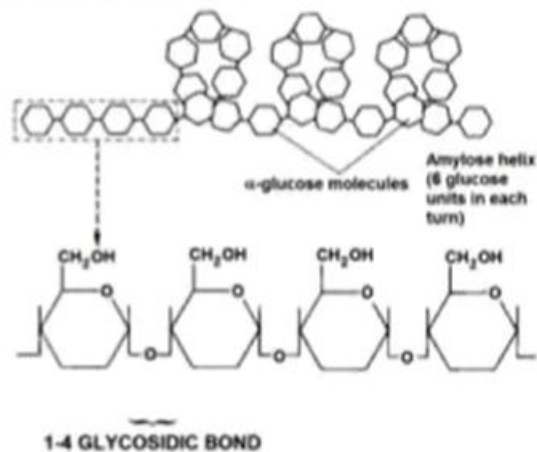
Condensation reaction



AMYLOSE

Stains deep blue with iodine
Relative molecular mass up to 50 000
Up to 300 glucose units/molecules
Unbranched helical chain

STRUCTURE OF MOLECULE

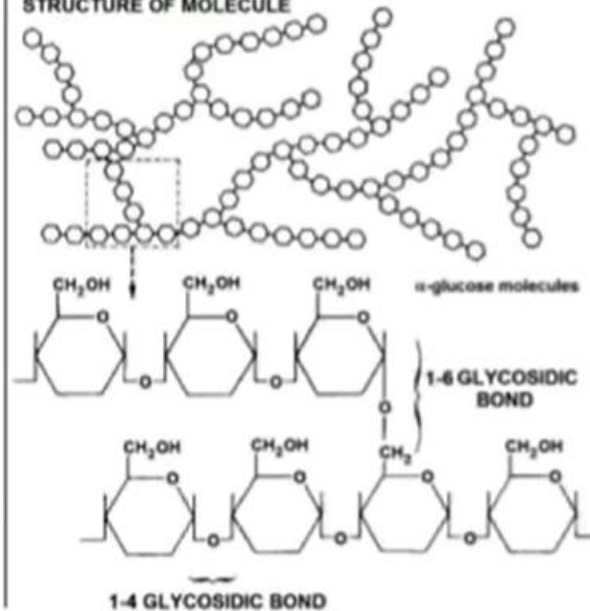


Linear molecule

AMYLOPECTIN

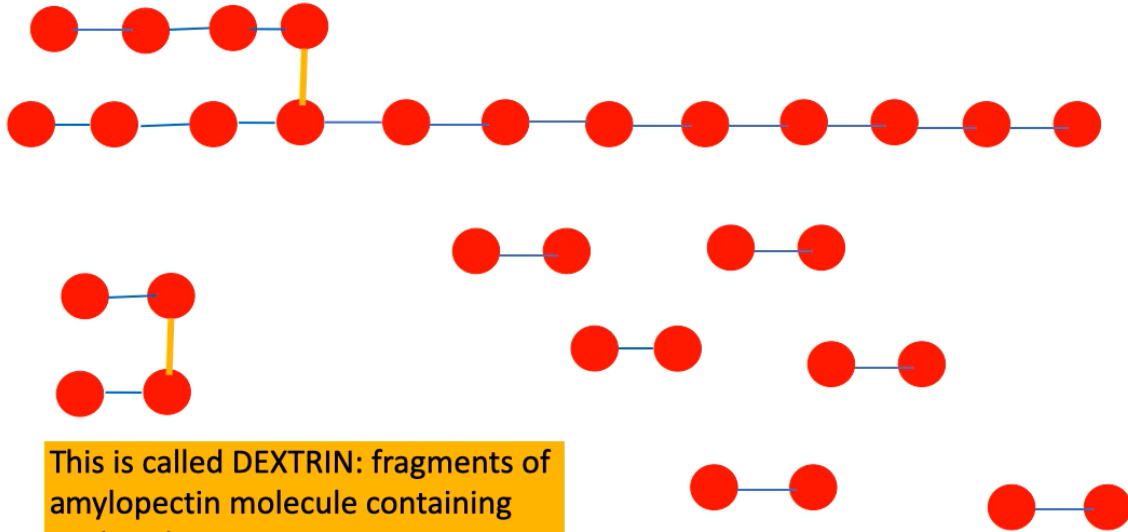
Stains red to purple with iodine
Relative molecular mass up to 500 000
1300-1500 glucose units/molecules
Branched chain

STRUCTURE OF MOLECULE

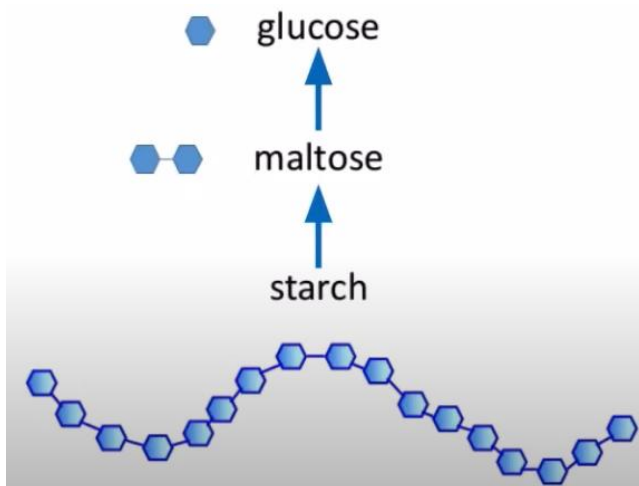


Branched molecule

Therefore, amylase cannot fully digest amylopectin



This is called DEXTRIN: fragments of amylopectin molecule containing 1,6 bond

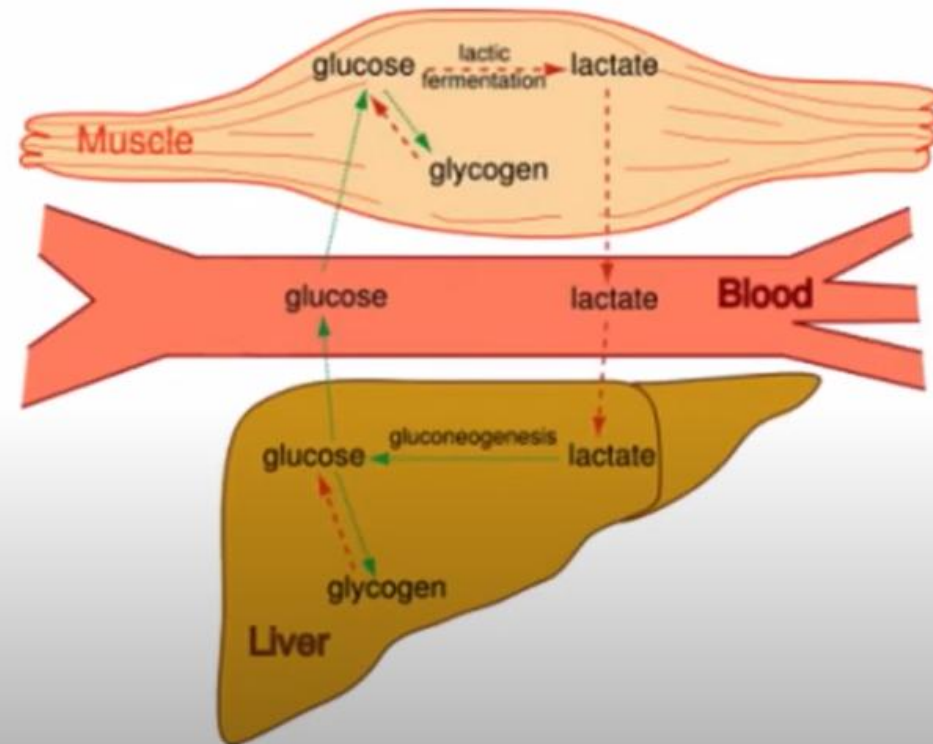


Glycogen - *function*

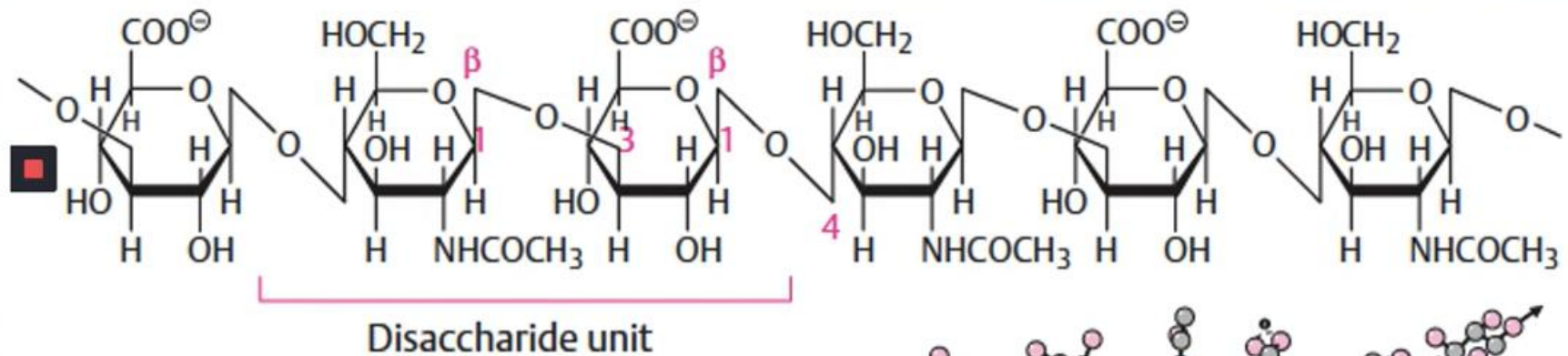
Glycogen is the storage form of carbohydrates in humans and other vertebrates.

When blood glucose levels *increase* (i.e. after eating), glucose is stored as glycogen in liver and muscle cells.

When blood glucose levels *decrease*, glycogen is broken down to release glucose into the blood so the cells can continue doing cellular respiration.

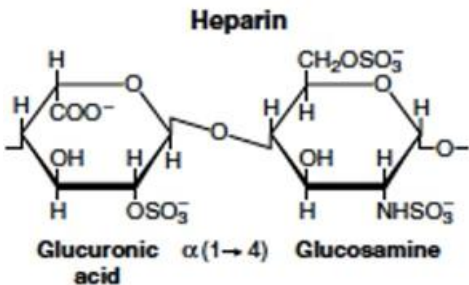
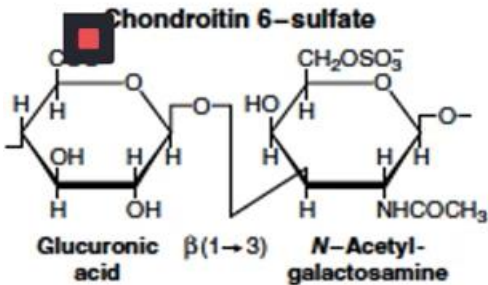
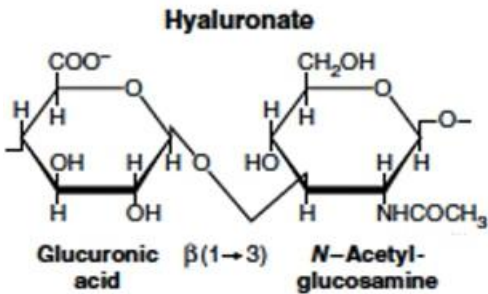


Glycosaminoglycans (GAGs)



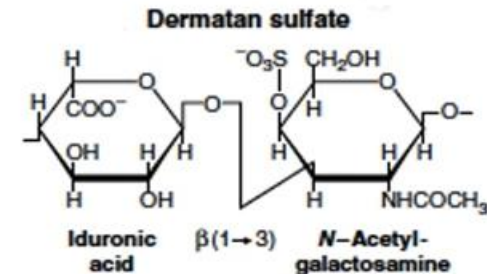
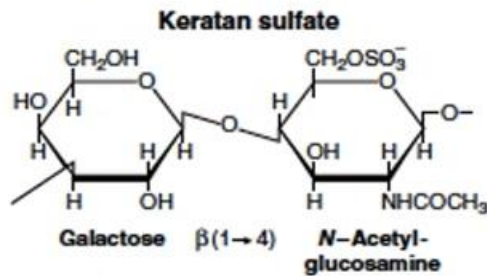
- a group of acidic heteropolysaccharides; important structural elements of the ECM
- made up of repeating disaccharide units, each of which consists of one uronic acid and one amino sugar





Glycosaminoglycans (GAGs)

We Can Understand!



- has repeating disaccharide units
 - uronic acid (glucuronic acid or iduronic acid)
 - amino sugar (*N*-acetylglucosamine or *N*-acetylgalactosamine)
- hyaluronic acid, chondroitin sulfate, heparin, keratan sulfate and dermatan sulfate

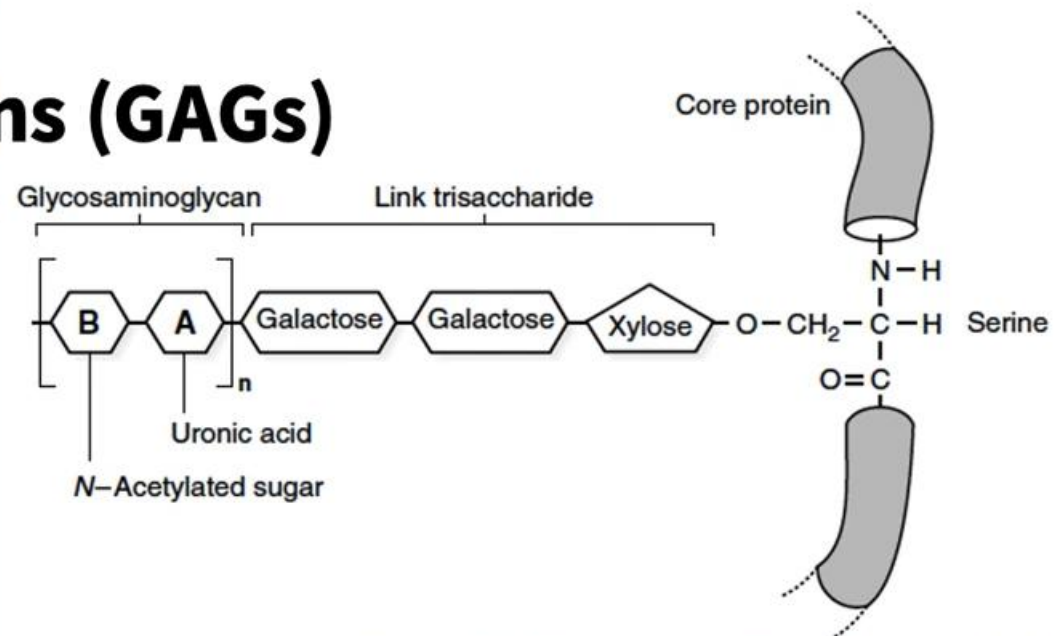


Glycosaminoglycans (GAGs)

Medical Biochemistry We Can Understand



Mark's Basic Medical Biochemistry A Clinical Approach, 5th ed Liberman and Marks



- least 6 types of GAGs
- except for hyaluronic acid, the GAGs are linked to proteins, usually attached covalently to serine or threonine residues
- keratan sulfate I is attached to asparagine
- attachment of GAGs to core proteins give rise to **proteoglycans**



Glycosaminoglycans and Proteoglycans

Medical Biochemistry We Can Understand!

Glycosaminoglycan	Function
Hyaluronic acid	Cell migration in: Embryogenesis Morphogenesis Wound healing
Chondroitin sulfate proteoglycans	Formation of bone, cartilage, cornea
Keratan sulfate proteoglycans	Transparency of cornea
Dermatan sulfate proteoglycans	Transparency of cornea
Heparin	Binds LDL to plasma walls Anticoagulant (binds antithrombin III) Causes release of lipoprotein lipase from capillary walls
Heparan sulfate (syndecan)	Component of skin fibroblasts and aortic wall; commonly found on cell surfaces

Glycoproteins and Proteoglycans

Medical Biochemistry We Can Understand!

• Glycoproteins

- • compounds containing carbohydrate, or glycan, covalently linked to protein
- carbohydrate may be in the form of a monosaccharide, disaccharides, oligosaccharides, polysaccharides, or their derivatives

• Proteoglycans

- subclass of glycoproteins in which the carbohydrate units are polysaccharides that contain amino sugars (glycosaminoglycans or GAGs)



Hexoses	Mannose (Man) Galactose (Gal)
Acetyl hexosamines	<i>N</i> -Acetylglucosamine (GlcNAc) <i>N</i> -Acetylgalactosamine (GalNAc)
Pentoses	Arabinose (Ara) Xylose (Xyl)
Methyl pentose	L-Fucose (Fuc; see Figure 13–15)
Sialic acids	<i>N</i> -Acyl derivatives of neuraminic acid, eg, <i>N</i> -acetylneuraminic acid (NeuAc; see Figure 13–16), the predominant sialic acid.

Glycoproteins (mucoproteins)

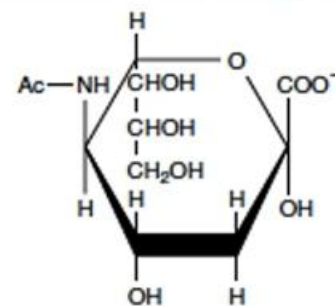
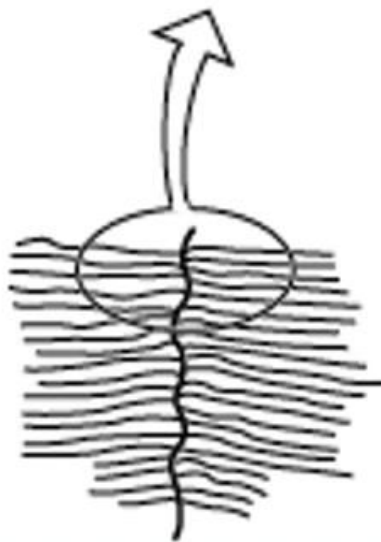
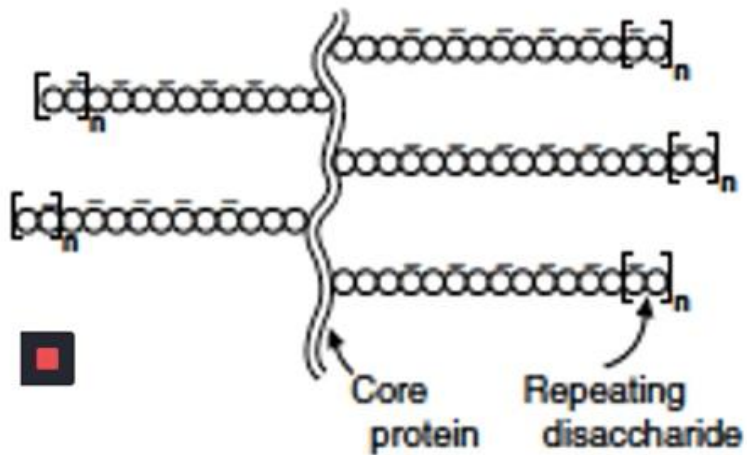


Figure 13–16. Structure of *N*-acetylneuraminic acid, a sialic acid (Ac = CH₃—CO—).

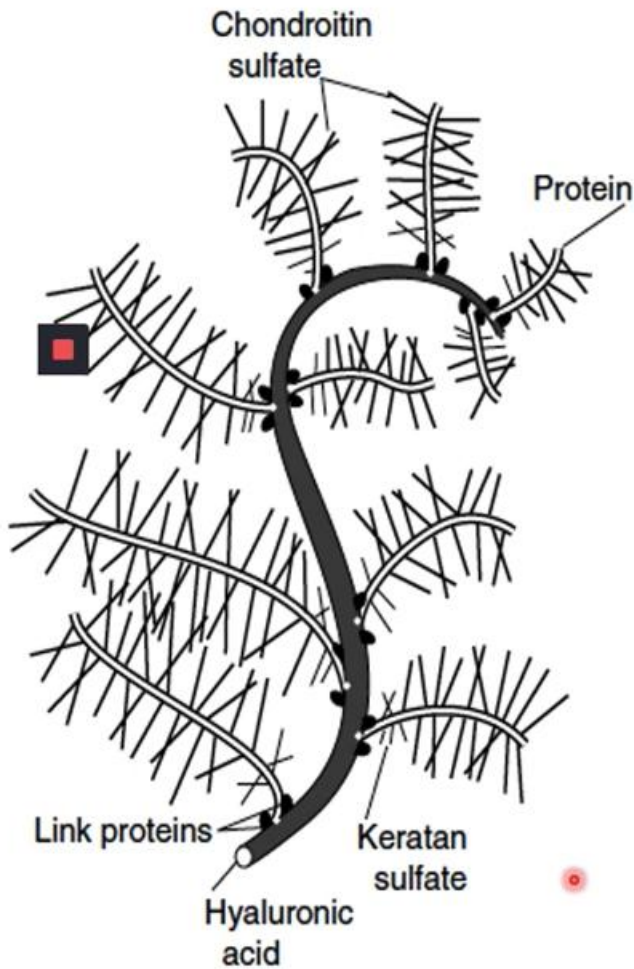
- in fluids and tissues, including the cell membranes
- proteins containing branched or unbranched oligosaccharide chains
- **sialic acids**
 - *N*- or *O*-acyl derivatives of neuraminic acid, a nine-carbon sugar derived from mannosamine (an epimer of glucosamine) and pyruvate
 - constituents of both glycoproteins and gangliosides



Proteoglycans

- contain many long unbranched GAGs attached to a core protein
- essential parts of the extracellular matrix, the aqueous humor of the eye, secretions of mucus-producing cells, and cartilage
- most GAGs have sulfated sugars contributing to negative charges → GAGs chains radiate out from the protein leading to overall “bottlebrush” structure
- may contain more than 100 GAG chains
- up to 95% oligosaccharide by weight





Proteoglycans

- possess negatively charged groups on the proteoglycan bind positively charged ions and form hydrogen bonds with trapped water molecules → hydrated gel
- provide the ground or packing substance of connective tissues; cushioning or lubricating functions



Mark's Basic Medical Biochemistry A Clinical Approach, 5th ed Liberman and Marks

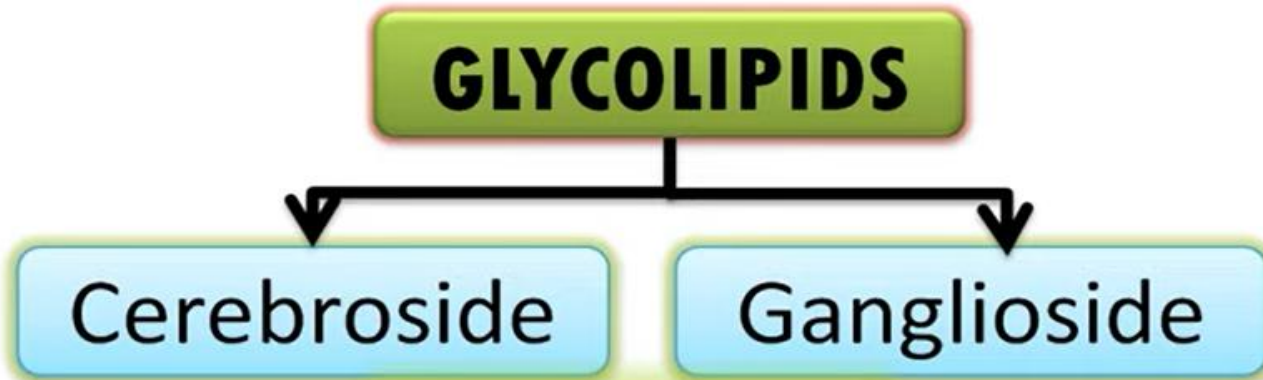
Hyaluronic acid, also called hyaluronan, is an anionic, nonsulfated glycosaminoglycan distributed widely throughout connective, epithelial, and neural tissues.

Mucopolysaccharidoses

Disease	Enzyme Deficiency	Accumulated Products
Hunter	Iduronate sulfatase	Heparan sulfate, Dermatan sulfate
Hurler + Scheie	α -L-Iduronidase	Heparan sulfate, Dermatan sulfate
■ Muroteaux-Lamy	N-Acetylgalactosamine sulfatase	Dermatan sulfate
Mucopolidosis VII	β -Glucuronidase	Heparan sulfate, Dermatan sulfate
Sanfilippo A	Heparan sulfamidase	Heparan sulfate
Sanfilippo B	N-Acetylglucosaminidase	Heparan sulfate
Sanfilippo D	N-Acetylglucosamine 6-sulfatase	Heparin sulfate

- caused by deficiencies of lysosomal glycosidases → partially degraded carbohydrates from proteoglycans, glycoproteins, glycolipids to accumulate within membrane-enclosed vesicles inside cells
- residual bodies can cause marked enlargement of the organ with impairment of its function
- affect multiple organ systems, with **bone** and **cartilage** being a primary target; significant neuronal involvement, leading to **mental retardation**

What are Glycolipids? Cerebroside vs Ganglioside



GLYCOLIPIDS

S
P
H
I
N
G
O
S
I
N
E

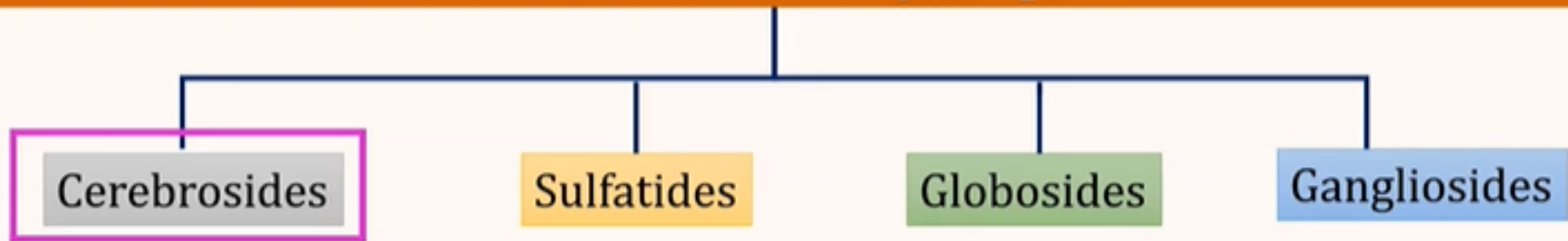
Fatty acids

Sugar

- Glycosphingolipids
- Important constituent of cell membrane and nervous tissues(brain)

- No glycerol
- No phosphate

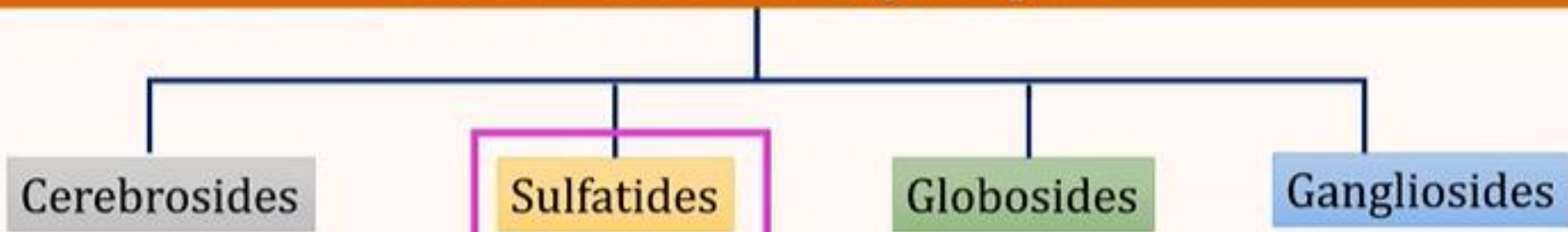
Classification of Glycolipids



Ceramide + Monosaccharide



Classification of Glycolipids



Ceramide
+
Monosaccharide
+
Sulphate

S
P
H
I
N
G
O
S
I
N
E

Fatty acids

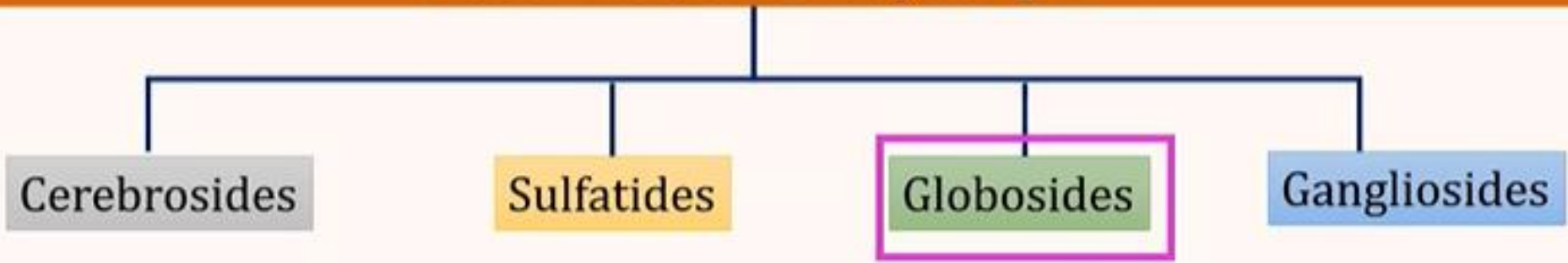
Monosaccharide

Sulphate

Sulfatides



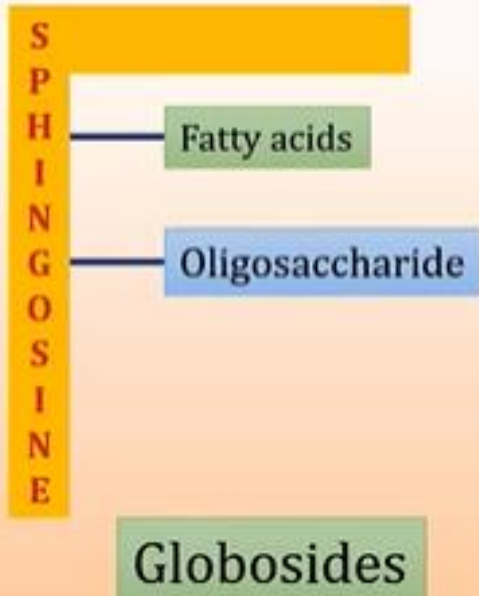
Classification of Glycolipid



Ceramide

+

Oligosaccharide



- Constituents of the RBC-membrane



Classification of Glycolipids

Cerebrosides

Sulfatides

Globosides

Gangliosides

Ceramide+ Oligosaccharide+
N acetylneuraminic acid(NANA)

S
P
H
I
N
G
O
S
I
N
E

Fatty acids

Glucose-Galactose

NANA

GM₃ Gangliosides

S
P
H
I
N
G
O
S
I
N
E

Fatty acids

Glucose-Galactose-GalNAc

NANA

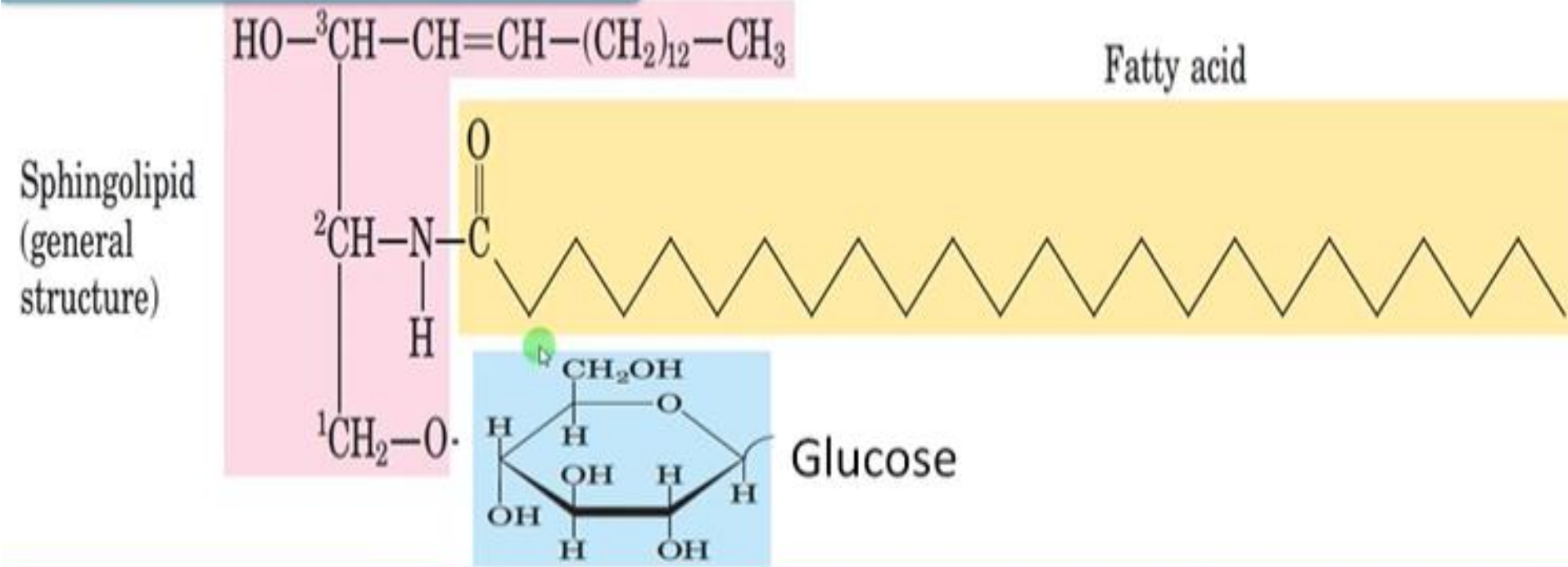
Galactose

GM₁ Gangliosides



Glycolipid (Glycosphingolipids) 1. Cerebroside

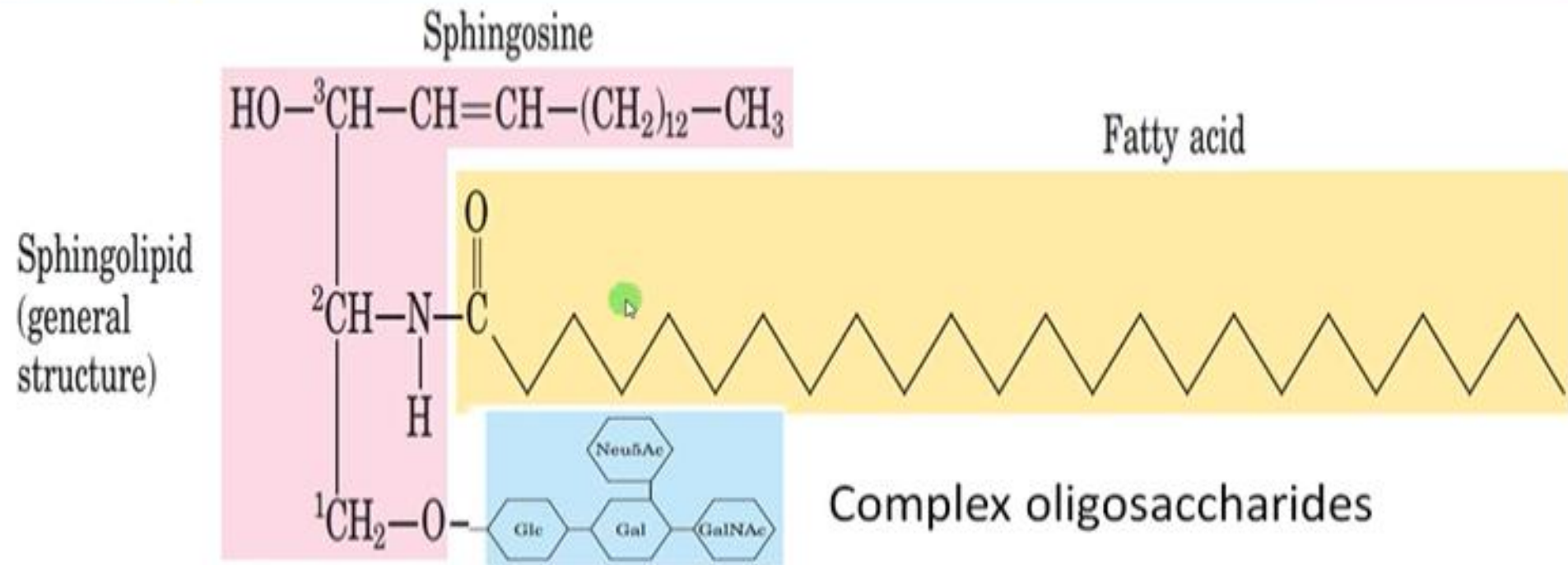
Sphingosine+fatty acid=Ceramide



Glyco: Sugar- lipid complexes

Formed by the joining of 1 or more monosaccharides connected directly to the $-\text{OH}$ at C-1 of the ceramide moiety by glycosidic bond

2. Ganglioside- Sialic acid-containing glycosphingolipids



Formed by the joining of complex oligosaccharides at C1 head group, (D-glucose, D-galactose, n-acetyl D galactosamine and N-acetylneuraminic acid (NANA) (sialic acid))