## III-S-(M.Sc-CBCS-Chem)-501-Photochem

## 2021

## Time : As in Programme

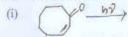
Full Marks: 50

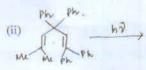
The figures in the right-hand margin indicate marks.

Answer all questions.

	(a)	Discuss the two types of transfer of excitation energy	1
•	(a)	with example.	
	(b)	Derive an equation to determine the rate constants	
		of photochemical reaction.	
	(c)	Write short notes on the following: 6	
		i. Phosphorescence	
		ii. Ferrioxalate actinometer	
		OR	
	(a)	Discuss the laws of photochemistry?	
	(b)	Write short notes on the following:	
		i) Chemical Laser	
		ii) Gas phase photolysis	
		iii) Chemical actinometry	
	(c)	Why only UV-visible region of light is able to produc chemical change, while others are not? 2	e

2. (a) Give the major product of the following photochemical reactions: 8





(b) Write short notes on the following:

8

- i) Di-ð-methane rearrangement
- ii) Photo reduction

OR

(a) Give the major product of the following photochemical reactions: 8

(b) Write short notes on the following:

- i) Norrish type-I reaction
- ii) Photo-Fries reaction of anilides
- 3. (a) Write down the major product of the following photochemical reactions: 8

- (b) Predict the product(s) formed and reaction mechanism on vapour phase photolysis of cyclohexanone?
- (c) Give the intermediates and final product of the following photochemical reaction.

tollowing photochemical reaction.

$$+ \frac{h^2}{M} \stackrel{\text{h}}{\longrightarrow} \stackrel{\text{h}$$