2021

Time : As in Programme

Full Marks: 50

The figures in the right-hand margin indicate marks.

Answer all questions.

(a)		ation 7
	(i) CIF ₃	
	(ii) AuCl ₄	
	(iii) BFCl ₂	
(b)	How do you represent the mode of bon	ding how
		6
(c)	How RRS is used to study the active of myog	lobin
* /	and hemoglobin.	4
(a)	Prove for NH ₃	6
	$\int vib=2A_1+2E$	
(b)	Explain basic principle of Resonance Ra	
	Spectra.	6
c)	What do you mean by reductant coordinate? V	Vhere
	and when does the redundancy occur in molecular	cules-
	explain.	. 5
	(b) (c)	in following molecules. (i) CIF ₃ (ii) AuCl ₄ (iii) BFCl ₂ (b) How do you represent the mode of bon ethylenediamine and diketonate complex and they are differentiated in IR spectra. (c) How RRS is used to study the active of myog and hemoglobin. Or (a) Prove for NH ₃ \[vib = 2A_1 + 2E \] (b) Explain basic principle of Resonance Ray Spectra. c) What do you mean by reductant coordinate? We and when does the redundancy occur in molecular to the coordinate of the c

	4.	(0	what is g value? calculate the g value (of T^{3+}
			Discuss the factors affecting g value.	:= x .:
		(b) Predict the ESR lines of following compoun	ds. 6
			(i) NH_2 .	
			(ii) ¹¹ BH ₃	
	*		(iii) $K_3[Mo(CN)_s]$	
		(c)		on in a
			magnetic field of strength 0.35T (g=2.0)	. 3
		(d)		2
			Or	
		(a)	Molecular formula of a compound is C_8H_6 ,	NMR
			spectra of such compound shows a single	et at
			2.98 δ (1H) and multiplet at 7.4 δ (5H). Prediction	ct the
			structure of the compound.	5
		(b)	of protons at a temperature of 25	°C is
			placed in a magnetic field of 2 Tesla, what i	s the
			ratio of number of proton spins in the lower sta	ite to
			the number in the upper state?	5
		5 %	$(g_N = 5.585, \mu_N = 5.05 \times 10^{-27} \text{ JT}^{-1})$	
		(c)	What are contact shift reagents? What is	the
		7.4V	significance of using these reagents in NMR?	4
7		(d)	What is the use of ¹⁹⁵ Pt NM R?	3
3.		(a)	Discuss the principle of Mossbauer spectrosc	ору
			of ${}^{57}Fe$ with the decay scheme	6

- (b) Predict the Mossbauer spectra of following compounds and write their structure. 6
 - (i) $Fe(CO)_5$.
 - (ii) $Fe_2(Co)_9$
 - (iii) $Fe_3(CO)_{12}$
- (c) Distinguish Sn^{2+} and Sn^{4+} , Fe^{2+} and Fe^{3+} from isomeric shift.
- (a) Write notes on 10
 - (i) Isomeric shift
 - (ii) Quadrupole interaction
- (b) What is the recoil energy of ${}^{57}Fe$ nucleus if energy of the emitted γ -rays from the first excited state is 14.4KeV?
- (c) The half life of the first excited state of ${}^{57}Fe$ is 1.5×10^{-7} S. What is the line width of the resonance line?