

I-S-(M.Sc.-Chem)-CBCS-406-  
(Spect-I)R&B

2019

Time : As in Programme

Full Marks : 50

Answer all questions. The figures in the right-hand margin indicate marks.

1. (a) Show that  $[a \cdot b \cdot c]^{-1} = c^{-1} \cdot b^{-1} \cdot a^{-1}$  where  $a$ ,  $b$  and  $c$  are group elements. 5
- (b) Write down the point groups for the following : 4  
Boat and chair form of Cyclohexane;  
staggered and eclipsed form of ethane.
- (c) Find out the representation matrices and representation of the bond vectors of  $\text{CHCl}_3$ . 5
- (d) Reduce the representation. 3

OR

(2)

2. (a) Discuss briefly the phenomena of absorption, emission and transmission. 6
- (b) What is the difference between natural line width and natural line broadening? 5
- (c) What do you mean by transition moment integral? What is its significance? 6
3. (a) What do you mean by Zeeman effect? Discuss its significance. 7
- (b) What is the term symbols for  $p^5$  and  $d^1$  configurations? 4
- (c) What is the difference between R-S coupling and spin-orbit coupling? 6

OR

4. (a) What do you mean by vibronic transition? How does a forbidden transition is allowed by vibronic coupling? 6
- (b) What do you mean by Franck-Condon active modes? Explain with examples. 5
- (c) How many electronic peaks are expected for  $d^1$  and  $d^2$  transition metal complexes? 6

(3)

5. (a) Write a note on photoelectric effect. 8  
(b) What chemical information is obtained from ESCA? 8

OR

6. (a) Discuss the classification of molecules according to their moments of inertia. 6  
(b) Discuss the effect of isotopic substitution and external field on the frequency and intensity of Microwave spectra. 5+5